

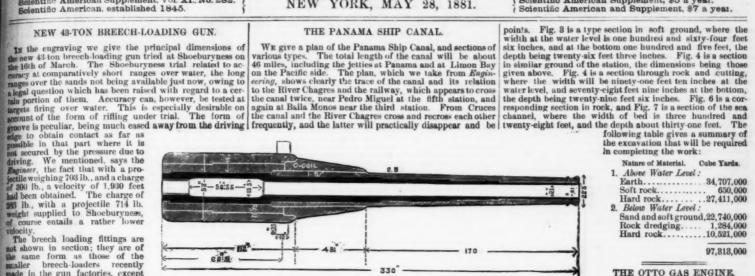
mtific American Supplement, Vol. XI., No. 282. mtific American, established 1845.

NEW YORK, MAY 28, 1881.

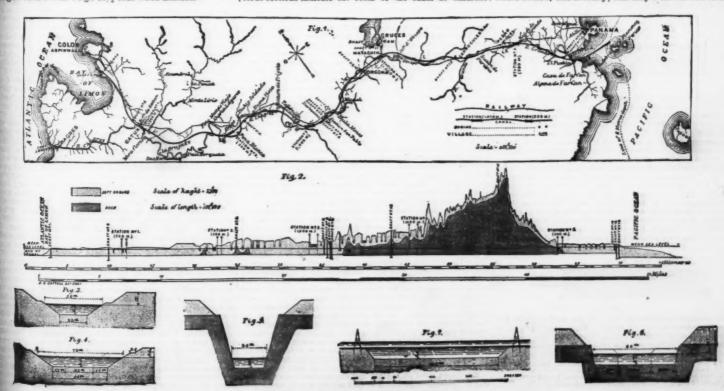
s Scientific American Supplement, \$5 a year. Scientific American and Supplement, \$7 a year.

Barth. 34,707,000
Soft rock. 650,000
Hard rock. 27,411,000
Below Water Level:
Sand and soft ground, 22,740,000
Rock dredging. 1,284,000
Hard rock. 10,521,000

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Station.	Approximate Distance from Colon.	Length.	Width,	Location.
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THE PANAMA SHIP CANAL

press a charge by the working piston in the working cylinder, and so render possible the combustion of comparatively

press a charge by the working piston in the working cylinder, and so render possible the combustion of comparatively dilute mixtures of gas and air.

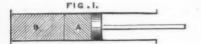
The plaintiff administered interrogatories to the defendant, Linford, who stated by his answer that it was not correct to say that in the gas motor made by him a charge of combustible and incombustible fluid was compressed by one instroke of the piston or otherwise, or that any such charge of combustible and incombustible fluid had been drawn into the cylinder by the previous outstroke of the piston or otherwise. It was the fact that in his engines the compressed charge, when ignited, propelled the piston during the next outstroke, and that the products of combustion were partly expelled from the cylinder by the next instroke. He admitted that his motor performed all the operations and effected all the purposes mentioned in the second claim of the plaintiff's specification, except as regarded the action of the slide, which in the plaintiff's patent admitted air alone during the first portion of the piston's stroke, and air and gas during the remaining portion, whereas in Linford's engine the slide valve only admitted a combustible charge of air and gas during the whole of such stroke. In such motors the piston was propelled by the explosion of the charge, and the products of combustion were expelled from the cylinder partly by the next instroke of the piston. A scavenger charge of air was afterwards drawn in to expel the remaining products of combustion. He referred to the specification of the manner in which his engine was constructed and worked.

Mr. Aston, Q. C., Mr. Hemming, Q. C., and Mr. Lawson,

charge of air was afterwards drawn in to expel the remaining products of combustion. He referred to the specification of his patent of 24th January, 1880, No. 330, for a description of the manner in which his engine was constructed and worked.

Mr. Aston, Q.C., Mr. Hemming, Q.C., and Mr. Lawson, were counsel for the plaintiff; and Mr. Kay, Q.C., Mr. Brett, and Mr. H. H. Cunningham were for the defendant.

Mr. Aston, Q.C., in opening the plaintiff's case, stated that although the gas motor engine was not a new thing at the date of the plaintiff's patent, yet the invention of the plaintiff enabled a very new engine to be employed much more generally and successfully than such machines ever were before. Mr. Aston then proceeded to explain at some length the principle of the first gas engine, in which a mixture of gas and air was exploded behind a piston in a cylinder. It was necessary to have a mixture of ordinary atmospheric air, so as to cause the gas, the carbureted hydrogen, to explode, by giving the particles of oxygen and hydrogen sufficient oxygen for them to form an explosive mixture. The proportions might be three to one of carbureted hydrogen, to six or seven or eight or ten of atmospheric air. If, into the charge chamber of an ordinary cylinder working an ordinary piston, there were introduced a combustible mixture consisting of eight parts atmospheric air and one of carbureted hydrogen, and if this charge were ignited by a small flame outside, an explosion would ensue which would drive the piston violently back from one end of the cylinder. If we had a similar chamber at the other end of the cylinder, we might repeat the process there, and drive the piston back again, and that would be the best form of an ordinary gas engine. Again, instead of using a second charge to blow the piston back, we might trust to the gradual cooling down of the remains of the exploded charge, when a partial vacuum would be made and the piston released, and this had also been done; and these were that the explosion produced a



chind that a charge of the combustible mixture of gas and tr. B (Fig. 1), and then by communicating as before the ght of a gas jet to the combustible mixture he would allow an mixture to be fired, causing the heat generated to be aborbed in a very large degree by the cushion of atmospheric ir, among which the particles of the combustible mixture could insinuate and disseminate themselves, and the operator instead of being sudden so as to cause violent shocks could be gradual, and economy would be effected by the ushion of air absorbing the heat, and expanding and doing lork. This, in its simple form, may be said to be a description of the important steps in advance made by Mr. Otto, and communicated to Mr. Abel in 1876. Mr. Aston then tent on to describe what took place in the engine after the tablosion of the charge and the making of a stroke. The ylinder would then be full of the products of combustion—ig. 2. There would be carbonic anhydride and the residue nd o



unconsumed atmospheric air. By the action of the fly-heel-the-piston would be brought back again, and as it ume back it would expel a certain portion of the products combustion. When, by the continued revolution of the y-wheel, the piston was again moved away from the closed ad of the cylinder, the residue was carried back as room

was left for it, and communication was opened to a supply, as provided by the patentee, of, first, atmospheric air, and then, behind, that combustible mixture; but the patentee provided for their being introduced in such proportions, and there was so much time given during the operation of the stroke of the piston, that instead of the three layers remaining divided and separate—as in Fig. 3—where A is air, B



combustible mixture, and C residual products of combustion—they became mixed together. The air and the residue of the charge commingled, the combustible mixture which was introduced last did the same, and the particles of the combustible mixture were dispersed through the entire charge; but they lay relatively more isolated, more dispersed, and more disseminated at the end next to the piston, less isolated, less dispersed, and less disseminated at the end next the point of ignition. By the action of the fly-wheel the piston now returned and compressed the clarge, and in that compressed state it was fired, and in that compressed state it was fired, and in that compressed state the same relative conditions of mixture remained unaltered, and as the combustible mixture had its particles nearest tegether close to the point of ignition, there was gradual combustion and development of heat and of force. Such, Mr. Aston explained, was the general principle of the action of the plaintiff's engine, but before going further he thought it would be well to say that the method of dealing with and introducing a charge, and of allowing that charge to be compressed, was one that had been proposed to be used in some specification prior to 1876; but, as far as was known, no gas engine was ever made and used which worked upon that principle.

Mr. Aston than read the specification of the plaintiff, commenting on it as he went. The first portion of the specification, apart from the drawings, we print, because on it turned a large part of the case. The drawings round which most interest concentrated, we give in full size from the specification further on.

"In gas motor engines as at present constructed an ex-

menting on it as he went. The first portion of the specification, apart from the drawings, we print, because on it turned a large part of the case. The drawings round which most interest concentrated, we give in full size from the specification further on.

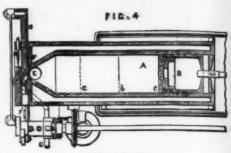
"In gas motor engines as at present constructed an explosive mixture of combustible gas and air is introduced into the engine cylinder where it is ignited, resulting in a sudden expansion of the gases and development of heat, a great portion of which is lost by absorption unless special provisions are made for allowing the gases to expand very rapidly. According to the present invention combustible mixture of gas or vapor and air is introduced into the cylinder together with air or other gas that may or may not support combustion in such a manner that the particles of the combustible mixture are more or less dispersed in an isolated condition in the air or other gas, so that on ignition, instead of an explosion ensuing, the flame will be communicated gradually from one combustible particle to another, thereby effecting a gradual development of heat and a corresponding gradual expansion of the gase, which, will enable the motive power so produced to be utilized in the most effective manner.

"The mode of using the gases and the arrangement of the engine may be variously modified in carrying out this invention. Thus, according to one arrangement the gases are introduced into the engine cylinder at atmospheric pressure. The cylinder is for this purpose provided with a slide having suitable ports for the admission of air and of an intimate mixture of combustible gas or vapor and air, and the movement of the silde is so regulated by means of a cam or eccentric on the engine shaft that during the first part of the stroke of the piston air alone enters the cylinder, while during a succeeding portion of the stroke the mixture of gas or petroleum vapor and air is introduced by means of a cam or eccentric on the engine may be affected with the condustion of the gases netroleum vapor and air is introduced, the particles of the mixture being situated nearest together at the positions dispersed in the air previously introduced, the particles of the mixture being situated nearest together at the position where they enter the cylinder and becoming gradually more dispersed as they mix with the air in front. A community of the combustible mixture is most dense, this ignites, and the combustible mixture is most dense, this ignites, and the combustible of the cylinder and the combustible of the cylinder and the combustible mixture is most dense, this ignite gradually as the flame extends to those particles that are more dispersed among the air. The gradual expansion of the gases thus produced causes the piston to complete its stroke, and on the return stroke, which may be effected either degrees thus produced causes the piston to complete its stroke, and on the return stroke, which may be effected either of combustible and incombustible fluid drawn into the cylinder of combustion are expelled through a valve, after which above-described, the gas and air being simply compressed to complete the cylinder of the cylinder is constructed of greater length than the stroke of the piston are charged of the cylinder of the cylinder is constructed of greater length than the stroke of the piston are cylinder on the cylinder is constructed of greater length than the stroke of the piston are cylinder in the confusion of the cylinder is constructed of greater length than the stroke of the piston are cylinder of the cylinder is constructed of greater length than the stroke of the piston of the gas and air being simply compressed to the cylinder is constructed of greater length than the stroke of the piston of the gase of the cylinder is constructed of greater length than the stroke of the piston of the greater length of the greater length of the cylinder is constructed of greater length than the stroke of the piston of the greater length of the cylinder is constructed of greater length than the str

worked well, but not so well as those described in the second part, beginning with the words, "according to another ort, beginning with the words, "according to another aringement." The patentee, according to one method, introduced air, and then behind it the combustible mixture with at compressing at all, and he fired them in that state. But it is all, "If you like you may use them in a compression,"

he sate, form."

Mr. Aston then proceeded to explain at much length how, by the use of a cylinder longer than the piston stroke, compression would be effected in the same cylinder in which combustion took place, as in Fig. 4, where A is the cylinder,



ABEL (OTTO), 1878.

B the piston, D the slide, E an exhaust port closed by a valve, not shown. When the piston is at the inner end of its stroke its face is at a; the slide D is in such a position that as the piston begins to move out, air entering by the aperture D, and port, C, until the piston reaches the point, b, when the slide reached such a point that gas is drawn in mixed with air until the piston reached the end of its outstroke; when the instroke was complete, and the compression effected, the slide moves so as to admit the gas flame, H. igniting the charge.

stroke; when the instroke was complete, and the compression effected, the slide moves so as to admit the gas flame, H, igniting the charge.

Mr. Aston laid much stress on the plaintiff's statement that the cylinder would be stratified. The work of compression was a very important function of the engine, for the proportions that constituted an explosive mixture were, say, 1 of gas to 8 of air at atmospheric pressure; but if we had a mixture in a compressed form, we might have a mixture in as high a proportion as 15 or 16 of air to 1 of gas.

Mr. Aston then, by means of a model, explained in detail to the Court the whole action of the Otto gas engine; and as this is no doubt fully understood by our readers who have followed us so far, it is unnecessary to say more on this point. Resuming, he called the attention of the Court to the statement in the specification, which runs: "It will be evident that if the space a', or a separate chamber, such as an air vessel," etc., what the patent then said was: "You need not always draw in when you make your first stroke, a charge consisting partly of air and partly of combustible mixture, provided that you have your chamber—which may be called the residuum chamber—sufficiently large to contain a charge of incombustible fluid, such as is left after the last charge has been fired, to act as a cushion in the way described."

Mr. Aston then defined the invention in the following

war Aston then defined the invention in the following words, as "consisting in introducing or admitting into the working cylinder a charge of a combustible mixture and an incombustible fluid in such a manner that the combustible mixture was dispersed in a gradually disseminated condition among the incombustible fluid, more dispersed at parts distant from, and less dispersed at parts near to, the place of ignition; thereby effecting, when the charge was fired, gradual combustion, gradual development of heat, and gradual expansion of the gases, and so utilizing most effectively the motive power, and avoiding shocks and waste of heat."

The claiming clauses, he submitted with the dead.

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The learned counsel illustrated this for the Court, by first putting into the cylinder of a model, white and afterwards red wool.

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MAY 28, 1891.

SCIENTIFIC AMERICAN SUPPLEMENT, No. 289.

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The frames and plating of the bottom to the upper part of the bilges to be thickly and efficiently covered with Portland cement, which may be mixed with sand in proportion, if cement, is sand. This must come even with all limber holes, which have to be cut in all floors at each side center line, and in intercostal plates between each frame space. Outside shell riveting to be cemented as usual before painting to make smooth work General Clause.—The whole of the material and workmanship to be strictly first-class; all defective material to be replaced, and all the work and material to pass the inspection of the superintendent (Mr. Samue! Holmes, New York), in accordance with the specifications.

the material and workmanship to be strictly first-class; all defective material to be replaced, and all the work and material to pass the inspection of the superintendent (Mr. Sanuel Holmes, New York), in accordance with the specifications.

Main Deck—to be of yellow pine 3x5 inches, to be well secured to deck beams by half-galvanized boils and nuts. The deck to be calked with three threads of oakum, and properly paid and made thoroughly water-tight. Margin plank, to be 9 inches wide, of yellow pine, thickness of deck. Main rail, to be 10x3/s inches, of oak, secured to angle iron on top of bulwark and to stanchions as described. Ceiling, to be 2½ inches in bottom, close ceiled to turn of bilge, and above this, battens 2 inches thick, spaced berth and strake. The ceiling in center to be in hatches so as to clean out the bilge. Windlass, to be a patent American Ship Windlass Co. windlass, for he 1 1-16 inch Stud cable chain with messenger chain attachment to be driven by froward hoisting engine. Pumps, two hold bilge pumps, to be fitted with brass chambers 6 inches diameter, fitted complete with brakes, pipes, cit., rendy for use; lead suction pipes into bilge, to be fitted; two composition fire engine and wash-deck pumps, to be fitted and so arranged that they can pump the bilge or draw from the sea, with proper stop cocks. The suction pipes to have a 8-way cock to arrange the work of pump. The fire and wash-deck pumps to each have 100 feet hose (rubber) and nezzles. Main house to be of iron. The coamings to be 12 inches high above the deck beams and 3½ thick, securely riveted to the tie plate by angle iron 3x8x3/s inches. Framing of house to be angle iron 2½x3½x5-16 inches, spaced about 24 inches apart, with as large knees as possible at top and bottom of 4-16 plate iron. Sides and ends of house 3-16 inch plate iron, flush butted, with but stray outside to form panels. Beams to be 25/x62x5 16 spread, one to each frame. Stringer to be fitted at sides 12x4-14. All to be single rivered, with 4/i inch rivets inishe

## IRON AND STEEL UNDER THE "HAY PROCESS

By A. T. HAY, of Burlington, Iowa.

By A. T. Hay, of Burlington, Iowa.

There is perhaps no subject in this, the iron age, within the wide range of practical engineering which has appealed to the earnest consideration of the scientist, metallurgist, and engineer more than that of the treachery of iron and steel in steam vessels, railway rails, bridge members, etc.

It is not strange, however, that wild theories, crude ideas, and hasty conclusions should find their way into public prints and scientific journals so long as there is no recognized science in what constitutes iron and steel, to predicate observations and experiments upon. And allow me to say, until there is a recognized science and an established constitutional make-up of the different varieties of iron and steel that is practically understood and appreciated by metallurgists, encineers, and consumers, there will be but little pecuniary benefit derived from or safety secured by random inspection or mechanical testing as regards stability, wherein they enter in land and marine construction.

Iron chemically pure is known to be a soft, weak, unstable, elementary metal, whereas homogeneous fine iron is practically an alloy containing several other rare elementary metals, including two or more metalloids chemically combined; while common coke iron and low grades of so-called Bessemer steel made therefrom are simply semi-mechanical mixtures of dissimilar stocks containing unreduced refractory oxides promiscuously combined with phosphorus, sulphur, sillcon, etc., including free and combined carbon, i. e., a heterogeneous amorphous mass of elementary substances. Now, these several unrevived refractory oxides are not only sources of mechanical season of themselves—like knots in lumber—but they are sources of chemical disintegration. Every one of these hard silicious knots is an electronegative center—a direct source of disintegration from voltaic action that is furthered by the corrosive elements or substances contained in water, including the crystallization of the iron from vibration, etc.,

Iron note chemical and mechanical causes.

Iron is not found in native masses to any extent (and then never chemically pure), but in a state of oxide, mineralized with other elementary metals and metalloids. And it is

upon these earthy so-called impurities, their proportion, electrical and pyro-electrical relation, process of reviving and manner of working, that depend the commercial value, adaptability, economy, and durability of any given product, that is, the adaptability of any piece of iron or steel for any given purpose depends entirely upon the number, kind, and relative proportion of the elementary rare metals entering into its constitution as an alloy in a metallic state.

It is upon this principle that we have Swedes iron, No-way iron, Russia iron, Scotch iron, Low Moor iron, Juniata iron, Brown's U. S. iron, etc. Each one of these several flue irons has a different constitution, a special adaptability and a world-wide reputation for its special superior quality that is derived from and assimilated with it by nature in the original ore peculiar to the locality wherein it is mined. It is these several rare metal alloys that constitute the blood, so to speak, that give to iron its true quality, valuable property, durability, etc., against chemical and mechanical influences. Such irons are isomorphous by nature and homogeneous in character. Whereas, a dissimilar mixture of different stocks of iron imperfectly revived, irregular in constituent elements, produce those beterogeneous, amorphous products known as common iron, or the so-called low steels that contain raw cinder or unrevived metallic oxides and other elements or factors of mechanical and chemical weakness, sources of deterioration and rapid decay.

With the foregoing premises I opened my investigation of iron and steel from an original standpoint that takes into consideration the chemical, electrical, and pyro-electrical relations of iron to all its mineralized elementary metals and metalloids with which it is found assimilated or combined within its native ore. In these find its constitution and a scientific solution of the whole question as pertains to its adaptation and stability.

Iron belongs to or is closely allied to a large family of earthy metals wi

phous metais play about as important a part of fine iron and steel as the vowels do in the make-up of fine of fine iron and steel as the vowels do in the make-up of fine of a double salt from any two metallic oxides of the same isomorphous group. This principle holds good in an isomorphous alloy with iron wherein the several metalloids will choose or combine with their first affinity analogous to the effect of manganese upon molten iron in the presence of oxygen. As is well known, the oxygen leaves the iron and combines with the manganese, and so it is with the several metalloids, neither will combine with iron in the presence of its isomorphous affinity. The best piece of iron to resist tension, vibration, corrosive and thermic influences, and at the same time remain malleable, ductile, and tenacious at all temperatures, will be that combination in alloy of isomorphous elements that will assimilate with or take up and utilize the greatest amount of sulphur, phosphorus, silicon, and other negative influences, leaving the iron the neutral or electro-negative metal in the alloy, i.e., complex iron is stable and reliable, clear iron unstable and treacherous.

plex iron is stable and reliable, clear iron unstable and treacherous.

Practical experience demonstrates that in about the ratio as the rare metallic oxides and the metalloids have been reduced to the minimum by the basic and other fluxing direct processes, the malleability and tenacity attained thereby is of a doubtful character, i. c., unreliable and treacherous in the extreme—instability seems to be inherent in all those so called mild steels, that consist in a low carburet of clear iron. The fact is, such metal has no constitution, and it stands as to genuine steel or fine iron about as bass wood stands to white oak for structural purposes. Some of these Siemens-Martin steels are as unreliable as untempered glass, and analogous thereto—will crack or split with a change of temperature or from the effects of cold water. I know of one railway corporation that had some thirty-five box plates break within a year, and nearly all of them gave way standing idle, dry, or when being filled up with cold water. But I have not found any such weakness where genuine crucible steel or double-worked fine charcoal homogeneous iron was used. Neutral ores containing the requisite elementary metals and metalloids assimilated in such proportions as will flux and revive themselves under the ordinary charcoal process into alloys, having the durability and valuable properties of fine iron or steel are very rare as commanded.

metals and metalloids assimilated in such proportions as will flux and revive themselves under the ordinary charcoal process into alloys, having the durability and valuable properties of fine iron or steel, are very rare as compared with the whole resource of iron.

A "red short" iron is one that contains pyro-electric elements: that is, an elementary metal or metalloid that changes polarization from an electro-positive relation to that of an electro-negative relation at a red heat when the oxygen and other negative elements present act on the iron. Carbon is also pyro-electric and changes polarization with iron several times between normal temperature and fluid heat. At about 700 deg. Fahr it is extremely positive when a piece of steel is flexible or very tough; at about 1,400 deg. Fahr, the carbon becomes intensely negative when a piece of steel is flexible or very tough; at about 1,400 deg. Fahr, the carbon becomes intensely negative when a piece of beated steel is at that temperature; the iron therein being the positive is "hot short" to friable, and at fluid heat the carbon again assumes the electro-positive and the iron becomes impervious to oxygen as is every day exemplified in the casting of steel and puddling of iron.

"Cold short" iron contains electro-negative elements that render the iron electro-positive at normal temperature, when phospborus that is ever present in such stock assumes the negative thereto. Now this so-called "cold short" iron may be changed to a malleable, duetile, fibrous iron by alloying it with electro-positive isomorphous metals sufficient to take the metalloid from the iron which gives to it a negative reaction. One of the finest, most ductile, and tenacious pieces of iron I ever saw was made from "cold short" stock alloyed with three rare elementary metals, and an assay showed as much phosphorus in the bar as in the original pig.

"Warm short" is another weakness known to certain

pig. ""Warm short" is another weakness known to certain grades of iron. It contains elements that are positive at normal that become negative toward iron at from 400 to 600 degs. Fahr. when the iron is extremely week, that change

to a positive and support the iron above 600 degs. Fahr, when it again toughens. Fine iron and true steel, as is well known, are easily injured or destroyed by overheating in a blacksmith fire. That is caused by the carbon present in the fuel being intensely negative in such temperature when the rare metals that are essential in fine iron and steel oxidize, which converts the one into brittle or burned iron, and the other into a brittle white carburet of iron. A demonstration of this fact may be had by restoring burned steel to its original fine quality through the agency of an isomorphous sponge containing the essential elementary metals of fine iron or steel, which may be added as a flux et a welding heat or in a molten state; hence, the theory that the carbon burns in overheated steel is not the fact, since steel may be restored without carbon.

To further demonstrate this, the finest grades of high steel may be successfully worked and welded at a yellow white heat by fluxing the metal with said isomorphous sponge, which absolutely refines and otherwise improves it; 3 per cent. of said isomorphous sponge added to Bessemer metal will set free or climinate the oxygen from the otherwise refractory oxides contained therein, and at the same time add an alloy of the essential elementary metals sufficient to convert the whole charge into a true steel, having all the valuable qualities and properties of crucible steel, that may safely carry 0.60 to 0.90 of 1 per cent. of carbon.

When homogeneous fine iron and true steel are polished and etched in an acid solution, the action will be uniform over the entire surface exposed, and if washed off in an alkali solution the oxidized surface will remain bright. Such metal contains the elements for its own preservation; whereas, when common iron or Bessemer steels are treated in like manner the action will be irregular, i. e., pitted and honeycombed or corroded in seams, and when washed off will turn black. Such an iron or steel contains the elements of its own decay. The ec

and steel depend entirely upon their constitutional or scientific make up for practical purposes.

Having from hypothetical induction and laboratory deduction demonstrated the essential constituent elementary metal contained in the different varieties of fine iron and true stee as also the allied elements that render iron hard, rigid, and brittle, including "red short" and "warm short" peculiar ties, and having tabulated them in the order of their seven electrical and chemical relations to ize brittle, including "red short" and "warm short" peculiarities, and having tabulated them in the order of their several electrical, pyro-electrical, and chemical relations to iron as regards oxygen, the metalloids, acids, and the alkalies, at normal and other temperature, and having previously discovered a complete and economic process of reviving the metals contained in refractory oxides in combination with iron, it was only necessary to formulate an "alloy sponge" to produce any given variety of fine iron or steel, and to apply the same as an auxiliary or secondary process to the knobbling fite or Bessemer converter to produce any desired quality of metal by eliminating the oxygen and thereby setting free the metals contained in the refractory oxides in stock, and at the same time adding such essential elementary metals necessary to produce any desired product. This investigation covers the best twenty years of my life. It is not a theory in an experimental state, but it stands forth as a verified, full-fledged scientific process. Every variety of fine iron known to the trade has been produced and repoduced from American stock. Varieties that will rival the finest Norway rod or the most durable Russia sheet have been made. Also a grade of metal hereofore unknown to the trade, having all the mechanical and chemical properties of fine iron and true steel combined, known as "Hay steel," that is particularly adapted for bridge members and other structural purposes, has been uniformly produced in large quantities.

The celebrated Glasgow, Missouri River, steel bridge,

other structural purposes, has been and other structural purposes, has been all steel bridge, The celebrated Glasgow, Missouri River, steel bridge, The celebrated Glasgow, Missouri River, steel bridge, and the first all steel truss bridge ever constructed, was the bridge of the first all steel truss bridge ever constructed, was the bright of the first all steel truss bridge. Iarge quantities.

The celebrated Glasgow, Missouri River, steel bridge, which is the first all steel truss bridge ever constructed, was built entirely of "Hay steel." The metal was furnished by Hussey. Howe & Co., Pittsburg, Pa. It was formulated by the inventor of the process, A. T. Hay, of Burlington, Iowa; blown, cast, and bloomed by the Edgar Thomson Steel Company, at Bessemer Station, Pa., under the immediate and personal supervision of C. Y. Wheeler, of the Hay Steel Company, Chicago. The shapes were rolled by Andrew Kloman & Co., Pittsburg. The bridge was constructed by the American Bridge Company, Chicago, under the general supervision of William Sooy Smith, chief engineer of the bridge, and projector of the enterprise in the interest of the Chicago and Alton Railroad Company, which was commenced, completed, and accepted within a year, and it was celebrated on the 6th day of June, 1879, by the good people of Glasgow, citizens of Missouri, and a delegation of distinguished American engineers. Since that date a swing bridge, on Kinsie street, Chicago, and a second gation of distinguished American engineers. Since that date a swing bridge, on Kinsie street, Chicago, and a second bridge across the Missouri River, at Plattsmouth, Neb., two spans 400 feet each, being of said material. Railway rails formulated and produced under said process, containing 0-50 of 1 per cent. Af carbon, are perfectly homogeneous and stand up under the heaviest traffic, and will carry over double the tonnage of common Bessemer metal.

### THE MANUFACTURE AND USES OF CAST STEEL.\*

THE MANUFACTURE AND USES OF CAST STEEL.\*

The first question which the user of cast steel has to answer is, to decide which of the three great methods of making steel produces a material best adapted to his own wants. Sir Henry suggested to you in his admirable lecture that Bessemer steel would answer every purpose for which steel is used, with the possible exception of the steel required to make Canadian axes. There can be no doubt whatever that the Bessemer steel of which Sir Henry gave you the first analysis would have made an excellent Canadian ax had it contained the proper quantity of combined carbon. The only doubt I feel is whether it could be produced of sufficient soundness without so large a percentage of waste as not to raise the price beyond that at which crucible cast steel for Canadian axes is now sold. I must confess that my experience of Bessemer cast steel would incline me to say that it could not. In spite of the prejudice that exists among consumers of steel, such is the competition of the present day, that I am sure that if steel could be produced of as good quality, and cheaper in price, by any other process than that of melting in crucibles, the present melting-furnaces of Sheffield would rapidly melt away into old bricks and mortar. I venture to express the opinion that the reason that high-class Bessemer steel is not as good as high-class crucible cast steel is because the former cannot be made sufficiently sound without the admixture of silicon and manganese, both of which substances are injurious to cast steel for most purposes. I fear that the advantages supposed to be derived from the use of manganese in the manufacture of cast steel are, to a large extent, illusory. I have frequently conversed with consumers of

\* Abstract of a recent lecture at Cutiers' Hall, London, by Henry St

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who knew the trade before the introduction of spiegel iron into Sheffield, and it is remarkable how many of them expressed the opinion that the crucible cast steel now in use is not as good as it was when they were young. To obtain sound ingots from high-class iron it is necessary to heil the steel for nearly half an hour after it has become faild, and then to allow it to cool down to a certain temperature before it is poured into the mould. The process is called "killing" the steel, and it is an axiom among them that the higher the quality of the steel the more "killing" it is also that the virtue of the process of crucible cast steel-melting that the virtue of the process consists; and the cost and quality of the cast steel produced depend in a large degree upon the skill brought to bear upon it. My theory is that the reason why high-class steel has to be so long solied is to get rid of its occluded gas, which would otherwise produce bubbles or "honeycombs" in its attempts to escape. The addition of a portion of scrap steel much assists the "killing," as would naturally be the case if we suppose the scrap, which has been melted before, to have parted with its occluded gas in the first melting. That the presence of manganese or silicon helps largely to "kill" the steel, I account for on the theory that the carbonic acid miltes with the manganese or silicon, and becomes a solid. So far my theory appears to hold water pretty well; but when I come to the fact that low-quality cast steel—for example, steel melted from Bessemer rail scrap, which contains from 0-15, to 0-05 per cent. of phosphorus—does not require any "killing" at all, and may be poured into the mould as hot as the strength of the crucible will allow, I am obliged to admit that I am not chemist enough to give you an explanation of the cause. The main point which I wish to impress upon you is that the much-maligned rule of humb, which insists upon the superiority of crucible cast steel over Bessemer steel for certain purposes, may have a scientific b

scientific basis, and must not be hastily set aside as prejudice.

Having decided by what process the steel is to be made, the next question that should come before the consumer of cast steel is the percentage of carbon which he wishes it to contain. When I first began business, the "temper" of steel, or the percentage of carbon which it contained, was concealed from the consumer. The following is a list of the most useful "tempers" of cast steel:

\*\*Rasor Temper\*\* (1½ per cent. carbon).—This steel is so easily burnt by being overheated that it can only be placed in the hands of a very skillful workman. When properly treated, it will do twice the work of ordinary tool steel for turning chilled rolls, etc.

\*\*Same File Temper\*\* (1% per cent. carbon).—This steel requires careful treatment; and although it will stand more fire than the preceding temper, should not be heated above a cherry red.

the preceding temper, should not be heated above a cherryred.

Tool Temper (1½ per cent. carbon).—The most useful
temper for turning tools, drills, and planing reachine tools
in the hands of ordinary workmen. It is possible to weld
cast steel of this temper, but not without care and skill.

Spindle Temper (1½ per cent. carbon).—A very useful
temper for mill-picks, circular cutters, very large turningtools, taps, screwing dies, etc. This temper requires considerable care in welding.

Ohisel Temper (1 per cent. carbon).—An extremely useful
temper, combining, as it does, great toughness in the
unbardened state with the capacity of hardening at a low
heat. It may also be welded without much difficulty. It is
consequently well adapted for tools, where the unhardened
part is required to stand the blow of a hammer without
snapping, but where a hard cutting edge is required, such as
cold chisels, hot salts, etc.

Set Temper (½ per cent. carbon).—This temper is adapted
for tools where the chief punishment is on the unhardened
part, such as cold sets, which have to stand the blows of a
very heavy hammer.

Set Temper (% per cent. carbon).—This temper is adapted for tools where the chief punishment is on the unhardened part, such as cold sets, which have to stand the blows of a very heavy hammer.

Die Temper (% per cent. carbon).—The most suitable temper for tools where the surface only is required to be hard, and where the capacity to stand great pressure is of importance, such as stamping or pressing dies, boiler-cups, etc. Both the last two tempers may be easily welded by a mechanic accustomed to weld cast steel.

Next to quality, by which is meant the percentage of phosphorus, sulphur, silicon, manganese, etc., the most important thing is temper, or percentage of carbon. For many purposes, indeed, temper is of more importance than quality. Nothing is more common than for steel to be rejected as bad in quality because it has been used for a purpose for which the temper was unsuitable.

When the steel has arrived in the user's hands, the first process which it undergoes is the forging it into the shape required. This process is really two processes. First, that of hearing to make it malleable, and second, that of hammering it, while it is hot, into the required shape. The golden rule in forging is to heat the steel as little as possible before it is forged, and to hammer it as much as possible in the process of forging. It is impossible to lay down exact rules for each of the thousand-and-one tools in which steel is used. The treatment of each tool in each process which it undergoes is an art that can only be learnt by practice, and can no more be taught in a lecture than the arts of skating, riding, or swimming. The utnost that can be done is to lay down certain general rules, and, if possible, to attempt some scientific explanation of them, to elevate them above the despised position of rules of thumb. The worst fault that can be committed is to overheat the steel. When steel is overheated it becomes coarse grained. Its silky texture is lost. If the temperature be raised above a certain point, the steel become

difficult. It must consequently be annealed. This process, like the preceding one, is a double process. The steel must be reheated as carefully as before, and afterwards cooled as slowly as possible. Many tools are only required to be hardened on a small part of their surface, and it is important that the unhardened parts should possess the maximum amount of toughness, the minimum amount of brittleness that can be attained. These tools must also be annealed after they are forged. The process of annealing, or slow cooling, leaves the steel coarse-grained, gives it its maximum of ductility, and causes it, in fact, to approach the properties of lead.

that can be attained. These tools must also be annealed after they are forged. The process of annealing, or slow cooling, leaves the steel coarse-grained, gives it its maximum of ductility, and causes it, in fact, to approach the properties of lead.

We now come to the culminating point in our manufacture, where the invaluable property which distinguishes steel from wrought iron or cast metal is revealed, a process by which we suddenly change our steel from lead into glass—the process of hardening. In this, as in all other processes which steel has to undergo, we have to run the gauntlet of fire. We do so, however, at greater risk than heretofore. The forging of our tool is finished; it has taken the final shape to which we have destined it, and whatever injury we inflict upon it by overheating is irrevocable, and can no more be cured or mitigated by the hammer. We must, therefore, double and red-uble our care, lest the temperature be raised above the point necessary to insure the required hardness. The part of the tool required to be hardened must be heated through, and heated evenly, but must on no account be overheated. Our tool must be finished at one blow—the blow caused by the sudden contraction of the steel produced by its sudden cooling in the water; and if this blow is not sufficient to give the steel a fine grain and silky texture; if, after the blow is given, the fracture, were it broken in the hardened part, should show a coarse grain or dull color, instead of fine grain and glossy luster, our tool is spoiled, and must be consigned to the limbs of "wasters." The special dangers to be avoided in hardening each kind of tool must be learnt by experience. Some tools will warp or "skeller," as we say in Yorkshire, if they are not plunged into the water like a knife, those of another shape must stab it like a dagger. Some tools must be hardened in a saturated solution of salt, the older the better, while others are best hardened, care should be taken to move it up and down in the water like a knife, those

a nother, the steel passes through the intermediate colors, of which the six above mentioned are arbitrarily selected as convenient stages.

It must be borne in mind that the elasticity of tempered steel is acquired at the expense of its hardness. It is supposed that the maximum of hardness and elasticity combined is obtained by tempering down to a straw color. In tempering steel regard must be had to the quality most essential in the special tool to be tempered; for example, a turning tool is required to be very hard, and is generally taken hot enough out of the water to temper itself down to a degree so slight that no perceptible color is apparent, while a spring is required to be very elastic, and may be tempered down to a blue. If you ask me to give you a scientific explanation of the process of tempering steel, I must confess my absolute ignorance. I have no more idea why it is so than the man in the moon, and the utmost I could do would be to mystify you in talking unintelligibly about molecular rearrangement and crystalline transformations. Hardening in oil is extent to attain by one process the change from lead into whalebone without passing through the intermediate glass stage, and is of great value for certain tools.

There are many kinds of steel to which your attention should be called, but which can only obtain from me the briefest mention. A special steel for taps, called mild-centered cast steel, is made by converting a cogged ingot of very mild cast steel, so that the additional carbon only penetrates a short distance. These bars are afterwards hammered or rolled down to the size required, and have the advantage of possessing a hard surface without losing the toughness of the mild center. Another special steel, somewhat analogous to mild-centered cast steel, is produced by melting a hard steel on to a slab of iron, or very mild steel heated hot enough to weld with the molten steel, so that a bar may be produced, one half of which is iron and the other half steel, or three-fourths iron and one-fo

skilled workmen it is useless. The addition of chromium instead of wolfram has somewhat the same effect.

It is much to be regretted that no easy method of testing cast steel has been invented. The amount of breaking strain and the extent of the contraction of the area of the fracture are all very well for steel which is not hardened and not required to be used in a hardened state, but for hardened and tempered steel it is practically useless. It is very difficult to harden and temper two pieces of steel to exactly the same degree. A single test is of comparatively small value, as a second-rate quality of steel may stand very well the first time of hardening, but deteriorates much more rapidly every time it is re-hardened, than is the case with high-quality steel. Nor am I at all sure that the breaking strain is a fair test of the quality of steel. For many tools the capacity to withstand a high amount of breaking strain slowly applied is not so much required as its capacity to withstand a sudden shock. The appearance of the fracture is very illusory. The fineness of the grain and the silkiness of the gloss is very captivating to the eye, but it can be produced by hammering cold. The consumer of steel may be enraptured, if he be of a poetical turn of mind, by the superb fracture of a bar of steel reminding him of a picture by Ruskin of the axis of the revolving hammer to the plane of the anvil. The practical consumer of steel must descend from the heights of art and science and take refuge in the common place of the rule of thumb, and buy the steel which his workmen tell him is full of "nature" and "body."

#### ENGLISH RAILWAY SPEEDS.

It may be interesting to some to have a table of the prin-ipal trains which average at least 49 miles an hour, start

to sto	D.	Dist.	Time.	Speed per hour.	
G. N.	Grantham to Kingscross	10514	128	49.3	
44	Finsbury to Peterboro'	78%	86	51.4	
66	Kingscross to Peterboro'	7614	90	50.8	
66	Hitchin to Peterboro'	4412	50	58-1	
6.6	Grantham to Peterboro'	20	34	51 3	
86	York to Grantham	88	100	498	
0.0	Grantham to Retford	831/4	89	51.1	
6.0	Retford to Doncaster,	8736	21	50.0	
-	(Summer Leeds Express, 33/4 h				
6.6	Kingscross to Grantham	10514	122	51.7	١
5.6	Grantham to Wakefield	70%	79	58.4	ı
M. R.	St. Pancras to Bedford	40 37	60	49.9	
66	Bedford to Leicester	4914-	60	49.2	
6.6	Kettering to Kentish Town	701/4-	85	49.4	
0.6	Manchester to Liverpool	34	40	51 0	
5.6	Bromsgrove to Cheltenham	811/4+	38	49.1	
C. L.	Liverpool to Warrington	15%	18	52.0	
N. E.	York to Newcastle	8314	102	49.0	
0.6	York to Darlington	4434	53	50.1	
G. W.	Paddington to Swindon	773%	87	58.8	
6.6	Swindon to Bath		36	49.2	
44	Oxford to Birmingham	6532	80	49.8	

### A CEMENT WANTED.

To the Editor of the Scientific American:

To the Editor of the Scientific American:

I would like to know if there is general need among amateur canoeists of a cement or calking substance to repair rents or holes punched by snags, cracks, etc.? If there is, I know of such an article. I am enthusiastically fond of canoeing, and have a light duck one, and would as soon think of going off on a cruise without provisions for the inner man as without this article. It is transparent. light amber color, readily applied, and ready for the water immediately after application, and calks perfectly and permanently. If there is no good preparation for this known to my fellow-amateurs, I will prepare a sample of it and forward it to you. If there is anything in use that will do as well, or well enough, I shall not trouble myself about; but, if not, I should consider it a duty to place it within reach of lovers of the delightful sport of canoeing—a sport which your journal has gone far to reuder general.

Mandatan M. H.

Manchester, N. H.

[We think there is need for a cement such as our corre-condent mentions.—Ep.]

### HOW SLUICE MINING ORIGINATED.

HOW SLUICE MINING ORIGINATED.

Col. Eddy, of this city, claims the credit of having originally introduced the sluice box for mining purposes, the invention owing its origin to an accidental discovery. He gives the following account of his connection with this important discovery: In the spring of 1850, when all operations were being carried on by the aid of the long tom and the rocker, he located a claim in the ravine just above the Catholic church in this city. There were several claims below him, the holders of which refused to permit him to run tailings on their ground. So he made a trough leading from his location through theirs and to a point below. On the bottom of the sluice, wherever the different sections joined, he nailed wooden cleats to keep the water and gravel from leaking through. At the lower end of the sluice he placed a rocker, and for one day manipulated the dirt that came down to it. At the end of the day he found that the rocker had saved scarcely any gold. Going along up the sluice, he found behind each of the cleats numerous sparkling particles of gold that had lodged there. He abandoned then commenced what he said was the first sluice mining ever carried on in California and probably in the world, so far as he knows. The sluice and riffles soon became popular, causing the price of lumber to advance rapidly. The Colonel says the only thing he regrets about his discovery is that he did not have it patented, and thus win fame and fortune.—Nevada Transcript.

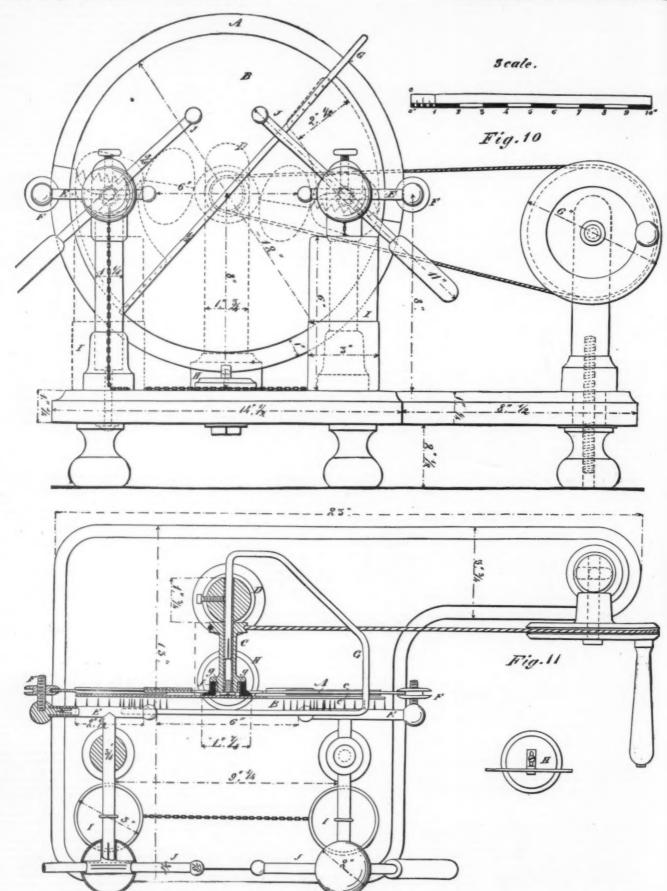
DETECTION OF ERGOT IN FLOUR.—The suspected sample is treated with cold ether or boiling alcohol to dissolve the greater part of the coloring matters of the flour. The residue is then extracted with ether, mixed with a small quantity of sulphuric acid, and the extract is examined with the spectroscope. The ethereal extract of ergot, if concentrated, absorbs all the refrangible portion of the spectrum beyond D; if the solution is diluted, the spectrum is enlarged, and there appear three absorption bands: the first between D and E, wave length 538; the second between E and F, wave length 499; and the third between F and G, wave length 467.

[Continued from SUPPLEMENT 279, page 4447.]

SIMPLE HOLTZ ELECTRICAL MACHINE.—CURATIVE APPLICATIONS OF STATIC ELECTRICITY.

By Geo. M. Hopkins.

Taken in connection with the articles in Supplements 278 and 279, the accompanying working drawings of a 12 inch Holtz electrical machine will be clear, and as all of the dimensions are marked on the drawing it will be unnecessary to go over the entire description in detail. Figure 1 is a front elevation, and figure 2 is a plan view. Of course



WORKING DRAWINGS FOR 12 INCH HOLTZ MACHINE

these dimensions need not be adhered to, but the proportions may be followed for machines larger or smaller than the one represented.

After what has already been said in regard to the construction of the Holtz machine the reader will have very little difficulty in understanding the double-plate machine shows in the annexed engraving, Fig. 8, which is a plan view giving only such parts as are necessary to convey an idea of the construction. In this view the parts described in connection with the single plate machine are designated by the same letters of reference.

An elastic rubber ring is placed between the centers of the plates and serves to communicate motion from the plate, B, to the plate, K. The required pressure on the rubber ring between the two plates is secured by turning a serew, O, in the end of the tube, M. By means of this arrangement the added plate is carried by the rubber ring when the machine is turned.

The combs. E, in this case are made double, as indicated in the engraving, the tube of which the comb is formed being bent into a U shape, and the points being inserted in the manner previously described. The central fixed plate is

The spark from this macaine will be no longer than that

881.

s plate s show , P, a

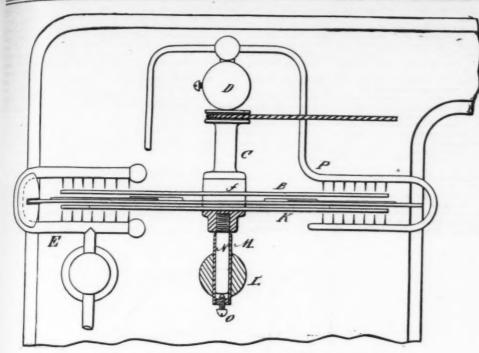
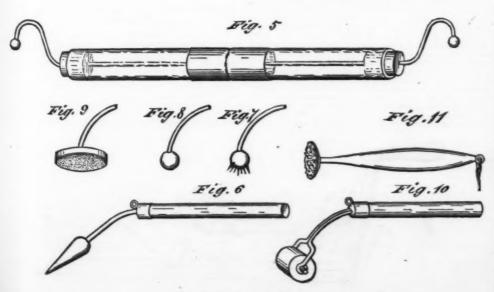


FIG. 3.—DOUBLE PLATE HOLTZ MACHINE.



Fig. 4—CURATIVE APPLICATION OF STATIC ELECTRICITY.



ELECTRODES.

of the single plate machine, but it will be "fatter," i. s., the quantity of electricity generated will be greater.

Should it be found desirable to perforate the revolving plates so as to place them on a single support or shaft, it may be proceeded with in the following way: A bole of about the size required in the glass is cut in a piece of board, and the latter is placed over the glass plate and held by weights. A copper or brass tube of about the size of the hole to be drilled is attached to a bow drill, breast drill, or bitstock, so that it may be revolved, and as it is revolved in the hole in the board, and in contact with the glass it is supplied with rather coarse emery and water. A little patience and a considerable care as the hole is nearly through, are all that is required to complete the job. Where one has a lathe the hole may be easily made with a corundum wheel of small diameter, revolved against the center of the glass and supplied with water. A sharp file, wet with turpentine, will speedily enlarge the hole to the required dimensions.

For curative applications of static electricity a double plate machine seems to be an absolute necessity, and one with four or more revolving plates is more effective than the simpler forms of machine, for, according to Dr. W. J. Morton, of New York city, a certain quantity is essential to success; intensity alone is insufficient. I have lately examined some very effective machines having four revolving plates, and constructed especially for the treatment of diseases, by Messrs. J. H. Berge & Co., 191 Greenwich street, New York city. These machines are furnished with electrodes of different forms for administering the current in different ways. The machines give currents that cannot be distinguished from the faradic by sensation experienced or by the effects produced. In addition to this they produce powerful shocks and other effects that are quite beyond the capabilities of ordinary battery appliances.

As before mentioned, a machine having at least two revolvin

revolving plates will be required to produce any beneficial effects.

Fig. 4 shows a method by which one may apply the electric current to himself, but it will doubtless be more convenient to have two assistants—one to turn the machine, the other to apply the electrode.

Fig. 5 shows a form of small double Leyden jar used in several sizes in connection with the machine for the purpose of condensing a limited amount of electricity in the machine. These jars are placed end to end, and lined and coated on the outside in the usual way. The curved rods are pointed on their inner ends, and extend nearly to the bottoms of the jars. The hooked ends of these rods are placed on the horizontal rods that support the collecting combs. The larger jars are removed, and the discharge rods are adjusted at a greater or less distance apart, according to the requirements of the case.

The patient sits in a chair upon an insulated platform, or as in the present case, with the legs of his chair in tumblers, which act as insulators. A board placed between the lower chair rounds forms a support for the feet, and is connected by a chain with one of the electrodes of the machine. By turning the machine one may charge himself with electricity, and if it is desired to treat any particular part, one of the electrodes shown in Figures 6 to 11 will be used, the glass insulating handle being grasped by the hand and the metallic portion connected with the ground, when the machine is turned and the electrode is presented to any portion of the body, sparks will leap from the body to the electrode.

The different forms of electrode are for producing different effects.

or otherwise connected with the ground, when he is turned and the electrode is presented to any portion of the body, sparks will leap from the body to the electrode.

The different forms of electrode are for producing different effects. Fig. 7 represents a pointed electrode made of either wood or metal, for drawing sparks from a local surface. Fig. 7 shows a ball with points for drawing the charge silently. Fig. 8 is a plain ball. Fig. 9 shows a wooden disk coated with metal filings and covered with cloth. Fig. 10 is a rolling electrode, and Fig. 11 shows a sponge electrode, such as is used in connection with faradic machines. The current adapted to this latter form of electrode is an induction current obtained from the outer coatings of the condensers. The method of controlling and utilizing the static induction current is the discovery of Dr. Morton.

Many other forms of electrode are used in connection with the machine, and insulated tables capable of supporting several patients at once are employed by physicians who have adopted "Franklinism" for the treatment of diseases.

### ELECTRICITY IN THE CARD-ROOM.

ELECTRICITY IN THE CARD-ROOM.

Or all the cares and trials of a carder's life, says the National Record, perhaps the one most mortifying is this electrical phenomenon. To enter the card room on a clear, dry, frosty morning in winter, or in the spring when a dry cutting wind is blowing, and see the fibers sticking and flying hither and thither, with "ends" out more or less on every condenser in the room, and apparently impossible to ever again get them to keep in place, is, to say the least, a tantalizing affair. Carders are not supposed to know a great deal about electricity, and generally the extent of their knowledge of it consists in sprinkling water on the rubs, or allowing steam to escape from a pipe provided for the purpose, and knowing that the effect is remedial for the time being. Why such an effect takes place, or the wherefore of this invisible demon of electricity, are alike subjects utterly beyond their ken. If it was a thing that could be fixed with a wrench, or even if it could be seen, then they could tackle it; but as it is, it proves a veritable imp of darkness and mischief.

It may be remarked that, so far as we know, it is in America only that carders have this trouble; and it is owing to the dry climate that such is the case, and not a peculiarity of the wool, as many suppose. In Europe they rarely or never see any manifestations of it in their mills, and we remember a story of an American geutleman who had lived on the borders of one of the New York lakes, where electrical conditions of the atmosphere were so strong that it was easy by rubbing the foot on the carpet of a room to draw a spark from the fingers sufficient to light a gas jet. He being on the continent, related this to some friends, who doubted the story, and he declared it was the easiest thing in the world, as he would show them; but judge of his surprise, when the gas refused to light in spite of his vigorous rubbing, much to the amusement of his friends. Electricians have often to keep heated irons around their apparatus wh

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property of attracting small bits of paper, straw, and the like; the vigor with which they will be attracted and repelled depending on the amount of friction to which the tube has been subjected. The tube has then become charged with what is termed electricity, and we say it has become electrified. Only that portion of the tube, however, to which the friction has been applied possesses this power of attraction; it has not diffused itself throughout the tube, therefore it (glass) is termed a non-conductor of electricity.

If we take a metal bar, say iron, and subject it to friction while being held in the hand, it will not be so effected, be cause, the human body being a good conductor as well as the iron bar, the electricity will pass through the body into the earth. But if the bar is held in a tube of glass, and then rubbed, the phenomenon will be manifested because it is then insulated through the non-conductibility of the glass. The electricity will, however, have diffused itself throughout the bar, and it will be manifested at the opposite end to that subjected to the friction. Therefore is iron termed a conductor of electricity.

The electricity will, however, have diffused itself throughout the bar, and it will be manifested at the opposite end to that subjected to the friction. Therefore is iron termed a conductor of electricity.

If a list of, say thirty, of the worst conductors were arranged in succession, in proportion to their lack of conducting power, then wool would come about in the middle of the list—after silk, which is a worse conductor than wool. Non-conductors are often termed insulators, and air, when very dry, is an excellent insulator: dampness, however, changing it into a conductor, as before explained. This shows us plainly why, in damp weather, we have no trouble in the card room with electricity, it is conveyed away by the atmosphere as fast as produced.

A dry, cold atmosphere is in the best form as an insulator or non-conductor; therefore do we on such days have trouble with the slubbings flying every way, clinging to the iron framework of the machinery, or to the person, when near enough, or any other conductor of the electric fluid.

It is clearly evident, from the foregoing, that it is static electricity, or that kind produced by friction, to be more plain, which is the source of difficulty, and from this it follows that the radical cure is to reduce the friction and moisten the atmosphere.

So long as the wool remains in a more or less bulky form, the phenomenon is not manifested, nor does it have to undergo such excessive friction until it is brought in contact with the rubs of the condenser, when, being at the same time divided into numerous fine filaments, it becomes overpowered by the electrical influence.

Another reason why the generating power becomes accelerated in cold weather arises from the fact that the oil is to a certain extent congealed in, and fails to effectually lubricate, the wool flows; that the greasy rub rolls are dry and rough, in the best condition for producing increased friction. The wool is also less elastic, supple, and altogether more liable to be excited by peculiar electrical con

to the same point as before the counteracting agents were applied.

When, therefore, the electricity begins to annoy you, I would advise the following plan: Take out all the rub rolls and scrape off the grease from each, afterwards rubbing with a clean oily rag to smoothness any roughness. Replace when clean and smooth, and set them apart one sixteenth of an inch, or at any rate so as not to touch in any part. Use a gauge, and set each end of each pair alike, commencing with the small stripper or wiper.

The roving does not require as much rubbing in cold as in warm weather, and the less it is rubbed the better, provided it comes off the spools well in spinning. Sprinkle a little oil on the side drawings, at the finisher feed rolls, previous to starting up, afterwards treating the rubs as described, which will in passing through further smooth them of any roughness that may have arisen from the scraping.

The trouble can also, in some cases, be immediately stopped by shortening the stroke of the eccentrics; but neither of these plans will entirely remedy the evil when considerable grease has accumulated on the rubs; the only thing to do in that case is to thoroughly cleanse them and make a new beginning. There is no need to go to extremes in setting them apart, so as to leave the rovings soft and flabby; the idea is to simply roll the rovings between the rubs without pressure, and this is all that is necessary under any conditions.

It may as well be stated here that the rubs are not intended aditio

without pressure, and this is all that is necessary under any conditions.

It may as well be stated here that the rubs are not intended to remedy defective carding, by attempting to rub inequalities out of uneven roving; they are there for the sole purpose of giving sufficient consistency to fine threads of wool, to enable them the more readily to unwind from each other in the spinning process; and all additional rubbing is unnecessary and mischievous. As a guide to others, the writer will mention that he has carded the finest quality of merino wool for fine flanuels, etc., in a room where for days together the temperature was not higher than 38° F., and this was done without any device for neutralizing electricity, as none manifested itself. Any one can do the same, in any kind of weather, by making intelligent use of the hints here given.

kind of weather, by making intelligent use of the hints here given.

Among the devices which have been tried as agents for neutralizing card room electricity, we may mention, that fine wire has been wrapped around the front condenser roll, wire of copper, of brass, tinned wire, iron wire, copper plated steel wire and annealed wire; and sometimes both front rolls have been covered. Rolls have been tried made of zinc, tin, copper, wood, glass, and hollow iron rolls, containing steam. Copper rolls set in front of the ordinary rubs are often employed, and are heated with steam; sometimes they are perforated, and are both made to revolve and to remain stationary. Wires stretched across for the rovings to touch, and then run into vrssels of water, have often been recommended; and we have seen a patented institution similar to a lightning conductor applied with as many points as there were slubbings and conducted with the metal conductor, which ran into water or into the ground. Steam pipes have been applied both to the rubs and independent of them, and to the feed rolls, breasts, and leaders in. The rubs have also been coated with various substances, as flour, ashes, and the like; so that there is no lack of devices to select from, and,

what is better, they are all sure cures, at once and for ever, etc., etc.

what is better, they are all sure cures, at once and for ever, etc., etc.

As an example of one of the above stated forms we will relate the contents of a letter, which gives a sure cure for electricity, as found in America, and then for comparison, another cure for the evil as found in Germany, which is also taken from a letter. The American letter instructs as

taken from a letter. The American letter instructs as follows:

"Take two wooden strips and tack them to the finisher frame, three inches from the rub roll, one on each side, and saw two slots in just opposite roving. Then take two bars out of the loom harness, place them in the slots on the under side of both top and bottom roving, letting the roving just touch them. Take a common small wire and loop it around each bar at one side of the card frame; let the wire be within one inch of the roving; let the lower end of the wire be six or eight inches long; lay a piece of iron on the card frame, bend the end of the wire so that it will be within one-eighth of an inch of the piece of iron, and you will have but little trouble with electricity. Carders, try it once."—Industrial Record, N. F.

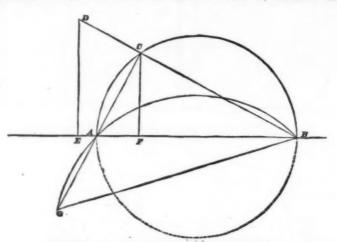
In Germany they go about it thus:

"The manufacturer in question found that during the process of carding his wool had got so electrified that it would not follow the narrow straps which in many continental condensers take the thread off the doffer. After careful ob-

realize these different conditions the slide, S<sub>1</sub>, is provided with a rounded groove which serves to guide it, and the slide, S<sub>2</sub> runs on the perfectly plane and polished sides of the frame. The space between b<sub>1</sub> and b<sub>2</sub> and the base may be filled with wedges, k<sub>1</sub> and k<sub>2</sub>, which are moved by the handle, b<sub>3</sub> of an eccentric. The relative positions of the different pasts of the apparatus having been perfectly adjusted at the beginning of the experiment, the block of cement to be tested is introduced beneath the pincers on the wooden support, u. The handle, b<sub>4</sub> is then turned, and the wedges at once place the slide, S<sub>3</sub>, in the direction of the axis of the apparatus. The screw, B<sub>4</sub> is now turned slightly, in order to fix the block of cement, and the support, u, and the wedges, k<sub>4</sub> are removed by maneuvering the handle, b. It only remains then to further revolve the winch of the screw. S<sub>4</sub> until the block of cement is broken as under. The apparatus recommends itself by its simplicity, its compactness, and its effectiveness.

#### THE QUADRATURE OF THE CIRCLE

A GEOMETRICAL construction of the quadrature of the circle, true to the sixth place of decimals, as a 3 in the sixth place of decimals is only an imaginary quantity, it is therefore absolutely correct in line drawing for all practical purposes. In fact, it is the only approximation to the quadrature of the circle I know of, that can be of any use whatever



QUADRATURE OF THE CIRCLE.

APPARATUS FOR TESTING THE TENSILE STRENGTH OF CEMENTS.

The apparatus herewith represented, one-eighth actual size, has been devised by Herr Kraft, of Vienna, for accrtaining the tensile strength of cements. The frame, B, which is bolted to the heavy wooden base, p, carries two slides, s, and s,. The block of cement to be tested is given the form represented at C, and its two extremities are grasped by the pincers, b, and b, the jaws of which are closed or opened by turning round-headed bolts located beneath the movable caps, r, and r,. Two other bolts under the caps, r, and r, allow the spring dynamometer to be determined when it is desired to make a trial. The dial of this dynamometer is provided with two index hands, the first of which, 2, turns in measure as the spring stretches under the action of the traction exerted, and when it has reached the point on the dial indicating a traction of 308 pounds, it carries along the second index, 2, by means of a tappet, d. When the brick of cement under trial breaks, the first index recoils and suddenly flies back to its initial position, while the other remains at the point which it had reached at the moment the rupture took place, and it is then easy to read on the dial the power that was exerted to sunder the cement. The slides are moved by turning the winch with which the screw 8, is provided. In order that the indications given by the apparatus as thus constructed be exact, the divisions in the dial must be absolutely equal, the tensile force must be exerted in the direction of the axis, and consequently the block must be placed in the axis of the apparatus, and, finally, the force exerted on the block must be transmitted integrally to the slide, S,. To

servations it was found that the cause lay in the material used for greasing the wool. When the oil was mixed with water and spirits of ammonia it became perceptibly electric especially with each change of the weather, and became such a nuisance that the spinning operation was seriously incurvenienced. This, however, only occurred with fine wool, which had been shorn unwashed, and then had been washed in a soda bath. Since the ammonia had been replaced by soda the inconvenience has almost disappeared, and it is only felt with great changes of weather." Deutsche Wolean of conjugation of ammonia cannot be disputed, and the cause of the phenomenon evidently lies here."—Textile Manufacture.

It is a true old saying that "too many cooks spoil the broth."

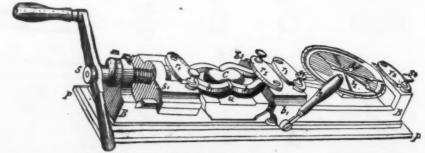
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# SCHOENBEIN'S OZONOMETER FOR FEBRUARY AND MARCH, 1881.

AND MARCH, 1881.

The curve of coloration for February shows an approximate coincidence of the color curve with the storm curve, but succeeding it in time, the synchrony appearing with exaggerated force on Feb. 28. The marked exception lies in the portions extending over the 5th, 6th, and 7th, where the high readings, alternating with low points, are not visibly connected with any atmospheric disturbances, the line of clear weather running uninterruptedly through this period. The changes of temperature seem to have influenced the test papers during these days, the deep tints appearing with the recurrence of warmer weather. On the morning



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of the 5th of February the thermometer indicated 10° Fah., which rapidly changed to 35° Fah. at noon. A similar change followed on the 6th and 7th, but the decline on the night of the 9th accompanied no such contrasted temperatures, both day and night being warm. It may be conceded that the coloration was heightened by moisture at every recurrent rise of temperature.

The threatening weather from the 9th to the 11th was coincident with storms westward, and the color curve reached its maximum for the month, doubtless influenced by the fog prevalent at that time.

The storm of the 13th, an easterly rain storm, was followed by clearing west winds and colder weather, and the color curve is maintained at 8 until the advent of the storm on the 15th and 16th, when it sinks to 0 and then rises with the weather curve.

on the 15th and 16th, when it sinks to 0 and their like weather curve.

Similar perturbations follow the stormy weather of the 16th and of the 18th and 19th, but it anticipates the weather ascension on the 20th, falling to 0 on the 21st, and again maintaining itself to the 26th through the influence of the stormy weather of the 23d and 25th, and the intermediate presence of a cold wave on the night of the 28d.

For March the same general coincidence strikes the eye, a succession of crests in each curve, occurring either nearly or quite together. When the color line rises elsewhere, as on the 15th and 16th, the 23d, 26th, and 27th, it seems due to wind, or at least on those days the wind was fresh and strong.

of the unal, does not admit of a doubt, but is a matter of fact and of truth. For it would seem to be a self-evident proposition that the first natural division of the unal is into halves, the second into quarters, the third into eightis. Consequently, the first natural scale of numbers is the dual or line scale; the second is the dual dual or surface scale; the third is the cubal or solid scale.

To illustrate: Take the line number, three, and cube it, and we find it is written 27 on the decimal scale. Now take the number six, and we find it is written 216. Now take twelve, and it is 1,728. Now twenty-four, and we find it to 13,824. Now let us see how these same numbers stand on the natural scale of cubes. The number three, cubed, would be written 33. The number six, would be 330. The number twelve, would be 3,900. Consequently, the conclusion seems inevitable, that all line numbers should be written on a line scale, all surface numbers on a surface scale, and cube or solid numbers on a cubal or solid scale.

The line number, three, written on a line scale would be 11. This number squared and put on a surface scale would be 21; same number cubed on a cube scale would be 33. Two feet in a perfect square. In presenting to you, to-day, 288 inches, or two square feet in a perfect square, I wish it to be distinctly understood that there are no pluses to be subtracted or minuses to be added, for if there was a small plus you might look for it in the center; on the contrary, you

"In a vat I place water and saturate it with any of the neutral saits or alkaline preparations by heating it with peroxide of manganese, muriate of soda, and sulphuric acid, in the proportions ordinarily used for generating chlorine, and when the chlorine ceases to come off from the saturated watery solution I then immerse the cordage and allow it to remain for a sufficiently long time to be perfectly saturated with the salts contained in the solution; it is then taken out and dried, when it is found to be uninflammable. To preserve this uninflammable condition the cordage is placed in a vat containing a solution of bichloride of mercury, to which is added a sufficient quantity of gelatinous matter to rerider it about the consistency of cream or thin varnish; after being thoroughly saturated with this solution the cordage is removed and dried."

# THE DENSITY AND TENSION OF SATURATED VAPORS.

THE DENSITY AND TENSION OF SATURATED VAPORS.

In a recent number of the Annalen der Physik, Herrn. Wällaer and Grotrian have described a careful investigation of the density and tension of saturated vapors, which they have carried out by well tested methods on the vapors of sulphide of carbon, chloroform, sulphuric ether, water, and account of the state of the question, and a brief summary of the results they have arrived at. The scope of their statement is as follows:

"Past experiments with regard to the density of saturated vapors have, with exception of Fairbairn and Tate, given values which little agree with those deduced, according to the mechanical theory of heat from Regnault's observations. The experiments of Herr Herwig, especially, give a considerably higher value for vapor-density than the theory, and also, for most liquids, a different increase of vapor-density with rising temperature. In the case of sulphide of carbon the ratio between the vapor-density calculated from the relation given by M. Herwig, and that obtained by theory, is nearly constant (the one is always about 4 per cent. greater than the other); but, on the other hand, for water and chloroform, the vapor-density increases considerably faster; for chloroform, e.g., the ratio at 30° C., is equal to 1043, at 100° it is 1 112. For water also at 100°, the ratio of the vapor-densities is already 1 111, whereas, at 11°, the vapor falls on Mariotte's law.

"Against the accuracy of the numbers got by Fairbairn

rolorm, the vapor-density increases considerably faster; for chloroform, e.g., the ratio at 30° C., is equal to 1.048, at 100° it is 1.112. For water also at 100°, the ratio of the vapor densities is already 1.111, whereas, at 11°, the vapor falls on Mariotte's law.

"Against the accuracy of the numbers got by Fairbairn and Tate, various objections based on the arrangement of the apparatus may be brought, and especially an exact determination of temperature is rendered very difficult.

"I'r Herwig's experiments have followed the density of the saturated vapors only to about atmospheric pressure, as the apparatus he used did not allow of compressing the vapor much more. He has thus left it doubtful whether the relation established by him holds good also for higher temperatures. In these experiments, too, a difficulty in determining the density of the saturated vapors appeared, in that the vapors, at least in part, were precipitated on the walls of the vessel before they showed the constant maximum tension. The precipitation should decrease with rising temperature. So that Herr Herwig assumes that, at higher temperatures, condensation of the vapor first occurs when maximum tension is reached.

"The question as to the density of saturated vapors is thus not settled experimentally, and we have, therefore, taken it up again, and sought to determine the vapor-densities for a series of liquids up to a pressure of about three to spheres. In this research it became necessary to constitute the walls. As it must be operative even before the deposit becomes visible, the density of the vapor must thereby come to be found too great. If we would ascribe the difference between Herr Herwig's values and those of theory to such an adhesion of the vapor, then the determination of the vapor-density in vessels of different size must yield different results, the density must be the less the larger the vessel. For the larger the wall-surface in proportion to the cubical capacity of the vessel, the greater must the fraction of vapor ac

vapor is independent of the size of the space in which is determined.

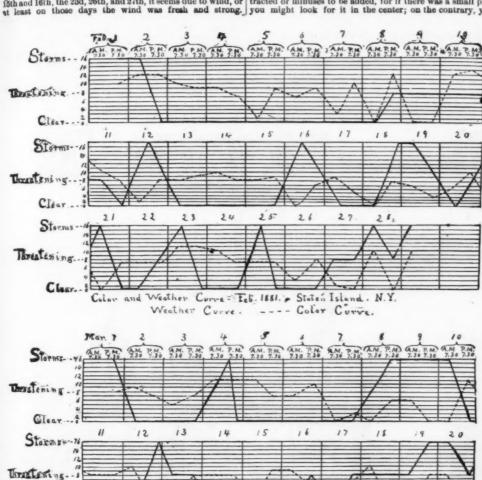
"On the other hand, the measurements confirm the result obtained by Herr Herwig, that vapors are precipitated before they have reached the so called maximum tension. They further prove that the tension at which condensation begins, the tension of condensation, stands in a relation of the maximum tension which is dependent on the nature of the liquid, but nearly independent of the temperature. The measurements appear to yield the unexpected result, that in general there is not a maximum tension in the hitherto accepted sense, but rather that the tension of saturated vapors, even when they are in contact with a large and excessive quantity of liquid, considerably increases through compression. It would appear as if the intermediate state, assumed by J. Thomson, were approximately realized. Any view of the process of vaporization must, therefore, be somewhat modified." (For details of the author's research we must refer to the original paper.)

### AURAL SYMPTOMS IN BRIGHT'S DISEASE.

AURAL STAPTOMS IN BRIGHT'S DISEASE.

CONTINUOUS or intermittent deafness has been described as a comparatively frequent concomitant of chronic nephritis by Rosenstein and Rayer. More recently Dieulafoy has called attention to such aural complications. Pain in the ears and tinnitus aurium have been added to the ist of significant precursors of uræmia by these authors. A separate thesis was lately published by Pissot, entitled "Lee troubles auditify dans te mad de Bright." Hitherto, however, the natury of aural complications in Bright's disease has been regarded rather in the light of functional disturbance than anatomical lesion.

Dr. Gurovitsch, of Odessa, records a case of parenchymatous nephritis where aural disease became a prominent complication, and where definite lesions were found. The case was that of a soldier, aged twenty-two, who had been a chronic sufferer from malarial affections. On his admission to hospital he had an intermittent fever of an irregular type. He complained of pain in the right ear and partial deafness. It was ascertained that he had parenchymatous nephritis



## SCHOENBEIN'S OZONOMETER

Weather Corve.

Curve. Mar. 1881

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and Weather

77th Street and 8th Avenue.

Storms -- 18

Threatening .

### THE NEW SYSTEM OF MATHEMATICS.\*

MATHEMATICS, like astronomy, was one of the first of the sciences to be investigated and, also, like astronomy, it is, as yet, but imperfectly understood. And however much we may admire the genius, the intellect, the wisdom of those great mathematicians of antiquity. Pythagoras, Euclid, and Archimedes, the truth is it were an impossibility for one man, or one era of great men 'o have measured the length, the breadth, the depth of this the science of sciences—this science without which the universe itself were an impossibility.

bility.

In confirmation of the position I have here taken, I have to say that the very basis of a true system of mathematics is the discovery of the natural scale of numbers.

Just when, where, or why, in the remote past, the so-called decimal system was first adopted is a matter of doubt and of speculation. But that the decem is an unnatural aggregation of unals, as the decimal is an unnatural division

\*Paper read by En Buldwin at the Union Meeting of the Mahoning and frambuli Teachers' Association, held at Niles, Onio, April 30, 1881.

Dr. Bauer, in 1877, showed the hygrometrical value of these papers, and though their color generally deepens, under conditions of humidity, some marked exceptions occur.

In April thallious hydrate tests and tests over water will be made.

L. P. Gratacap.

L. P. Gratacap.

Staten Island . N. Y.

Color Curve.

30

### UNINFLAMMABLE FABRICS.

To the Editor of the Scientific America

To the Editor of the Scientific American:

I see by your Supplement, page 4221 (January 29, 1881), you publish an item. "That the French Society for the Encouragement of National Industry have awarded to M. Martin. of Paris, a prize of 1,000 francs for his preparation rendering textile fabrics uninfiammable." It is announced as a new discovery; and, as I am not willing the credit should go to France, when the United States is entitled to it. I would respectfully call your attention to the fact that the same was pattented on the application of my brother, James H. Johnson, M. D., as early as 1850, and the patent granted to me as his administrator by our government, under date September 26, 1850.

Owing to the death of my brother its manufacturing was abondoned, and nothing done with it since.

Galena, Ill., 1881.

The Johnson patent referred to by our correspondent,

The Johnson patent referred to by our correspondent, ranted September, 26 1850, was for rendering cordage uninammable. The following is the process as described in the pecification of the patent:

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Soon a purulent otitis media was developed, and later the previously healthy left ear likewise became implicated. Facial oderna was superanded to the other symptoms, and this became more marked with the increased violence of the aural symptoms. When the latter showed an amelioration, the oderna also partially subsided. Gurovitsch is at a loss how to account for this coincidence. The patient died of pericardial dropsy and cardiac failure, and at the autopsy the aural lesions were found to correspond to the diagnosis of otitis media purulenta.—Berl. klin. Woch.

#### THE CHIME AT ST. GERMAIN L'AUXERROIS.

FORMERLY when church chimes were rung by hand, it as always effected by maneuvering the clappers inside the bells by the aid of pedals, and of keys arranged on a

was always effected by maneuvering the chappers inside the bells by the aid of pedals, and of keys arranged on a sort of key board.

The old automatic system consisted of a metallic cylinder covered with numerous projecting pins, and rotated by a clockwork movement. In revolving, this cylinder raised, by means of the pins on its surface, the levers, which were connected by wires with the chappers; and the latter, on falling back against the bells, produced the tones necessary to play the desired airs.

When it was desired to obtain very powerful tones and to play tunes of some length, an enormous power had to be expended. By the aid of the key board and its attachments it was necessary, in fact, to raise chappers, which in some cases were extremely heavy, and to overcome a resistance that could be surmounted only by great trouble, and that too by striking with the fists on the keys and with the feet on the pedals. (Fig. 1.) When this automatic cylinder was employed, it was necessary (owing to the fact that it had to raise heavy clappers) to give it dimensions varying

day. Fig. 2 shows the chime as now definitely located in the tower of St. Germain l'Auxerrois.

day. Fig. 2 shows the chime as now definitely located in the tower of St. Germain l'Auxerrois.

In the old systems of chimes there was but one mechanism for all the bells, but in this each bell has a special set of motive wheels of a power proportioned to the weights. As each set of wheels can operate only when it is desired to obtain the sound of the bell connected with it, it uses its power only when it acts. The manufacturer has based his calculations of the motive weights on data from experiments which have shown that, in large bells giving the bass tones in the chimes, the effort necessary to raise the clappers is about equal to a hundredth of the weight of the bells; a bell of 4,000 pounds, for example, requiring an effort of 40 pounds. But in the small bells the effort is relatively greater in proportion to their weight, and may be estimated at about one-fifth; so that for a bell of 20 pounds, for instance, an effort of 4 pounds would be necessary. As may be seen, then, this division of the weights and sets of wheels affords considerable economy in motive power. This arrangement, moreover, has the advantage that the wheels operate with sufficient speed to allow passages to be readily played which contain quavers and semiquavers.

Finally, it may be remarked that one of the characteristic features of this chime is that the operations of the automatic cylinder are effected without any outside aid—the mechanism once wound up taking care of itself.

cocoons alive from abroad; the next great difficulty is the struggle against the climate, which has been my greatest enemy here during the last two years. Artificial hest, unless pure air and a free ventilation can be obtained at the same time, cannot replace natural heat. The reverses experienced during the last two years in my a'tempts to reproduce and rear these splendid silkworms and other Lepidoptera have been beneficial to me in one respect—they have given me valuable information, which I should never have acquired had everything succeeded according to my wishes, although I sincerely hope these fatalities will not occur again, at least on such a large scale.

The species of silkworms which I placed on trees in my garden during the magnificent month of August were Cynthia, Pernyi, Prometheus, Cecropia, Polyphemus, Luna, and Pyri.

The Alachic worms thrived remarkably well on the Alachic The Charthic worms thrived remarkably well on the Alachic The Charthic worms thrived remarkably well on the Alachic The Charthic worms thrived remarkably well on the Alachic The Charthic worms thrived remarkably well on the Alachic The Charthic worms thrived remarkably well on the Alachic treatment of the contraction of th

greater in proportion to their weight, and may be estimated at about one-fifth; so that for a bell of 20 pounds, for instance, an effort of 4 pounds would be necessary. As may be seen, then, this division of the weights and sets of wheels affords considerable economy in motive power. This arrangement, moreover, has the advantage that the wheels operate with sufficient speed to allow passages to be readily played which contain quavers and semiquavers. Finally, it may be remarked that one of the characteristic features of this chime is that the operations of the automatic cylinder are effected without any outside aid—the mechanism once wound up taking care of itself.

SILK-PRODUCING BOMBYCES REARED IN 1880. By Alfred Wallly, Membre-Laureat de la Societe d'Acclimatation de France.

As it has been seen in my report, published in the Journal of the Society of Arts, February 13 and March 5, 1880, the cold weather in 1879 had the most disastrous effect on the

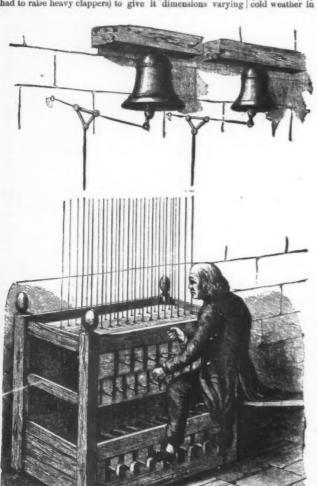


Fig. 1.—KEY-BOARD CHIME.—OLD SYSTEM.

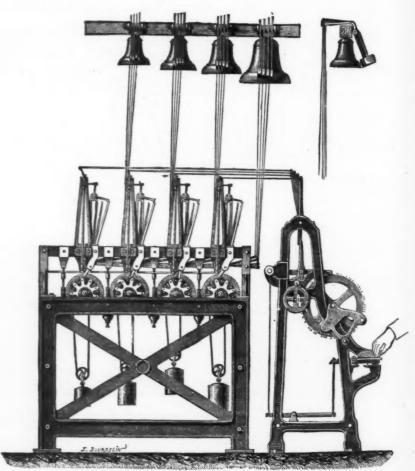


Fig. 2.—NEW CHIME OF SAINT GERMAIN L'AUXERROIS.

from \$\frac{1}{4}\$ to \$\frac{1}{4}\$ feet in diameter, and a very considerable rotary power. To actuate these machines, weights of from 1,000 to 6,000 pounds were employed. These were suspended from chains rolling around drums, and were wound up by means of windlasses, requiring the labor of from one to three men for one or two hours. The chime of the Samarit sine, at Paris (destroyed in 1813), was on this system. At Bruges, Belgium there is a very perfect chime of the same kind. It is provided with a bronze cylinder weighing no less than \$22,000 pounds, and representing a value of \$\frac{1}{8}\$,000 for the metal alone. These old-fashioned chimes, as may well be seen, were very costly and inconvenient. They utilized at the very most only 30 per cent., or at the least, one third of the force expended. This will be readily understood if it be taken into consideration that the cylinder, always revolving with the same speed, expends as much power to sound a 6-pound bell as to strike a large one weighing 6,000 pounds.

The chime recently placed in the Saint Germain l'Auxerrois, at Paris, differs essentially from those above described and marks a great progress in the manufacture of these apparatus. It consists essentially of: (1) sets of wheels and weights equal in number to that of the bells and of power proportionate to the weight of the latter; (2) of a key board similar to that of a piano and operated the same way; and (3) of a cylinder for playing automatically, of about 16 inches diameter, so arranged that it can be removed and replaced by others affording other tunes. The importance of the revolution effected in the mechanism may be realized when we state that this chime, instead of needing the labor of two or three men during a part of a day for winding it up, requires but the labor of one man for ten minutes each week to wind up the weights connected with the automatic part, which is arranged to play four airs per

rearing of exotic Lepidoptera. In 1880, the fine and warm weather we had during the month of August and part of September allowed of the successful rearing of several species in the open air; but the cold and wet weather, lasting of several species in the open air; but the cold and wet weather, lasting of several species in the open air; but the cold and wet weather, lasting of several species, as in 1879. The moths of 8 Promethea, for instance, which generally emerge at the end of June and beginning of July, had not all emerged before the end of August; one, 8. Oynthia. emerged on the 7th of September. Excepting a few, the Indian species did not emerge at all. Four Actions from the paps state. No pairing could be obtained. The same failure attended Attaeus mylitia (Himalaya race), of which occoons placed in a hot house at a gardener's, in April, did not produce any moths till August, too late to rear this species. If fertile ova could have been obtained.

At the end of February, 1880, I received from Calcutta 900 cocoons of Attaeus mylitia (Himalaya race), about 750 of which had died in transit, but none had emerged. Laters on April 19, I received from Major Coussmaker a tin box containing 100 Mylitia cocoons of the Bombay race, more than two-thirds of which had emerged on the way. The tin box had evidently been kept in too warm a place, and the melia of May to the beginning of July, but, as with the Himalaya race, no pairings took place, and the melia of the cocoons of Attaeus mylitias of which had emerged on the way. The tin box had evidently been kept in too warm a place, and the melia forms and those of the cocoons of July, but, as with the Himalaya race, no pairings took place, and the melia of the produce and the produce

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MAY 28, 1881.

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Attaeus pernyi (North China).—This most valuable oak silkworm, now thoroughly acclimatized in Spain, where it is cocoons I sent to various parts. A correspondent in Illinois writes that it was double-brooded there, and that he found some of the worms (which had become dry and tough in consequence of the hot, dry summer), feeding on hawthorn bushes growing close to the oak trees. Other Pernyi larvæ were found on appletrees in a garden, where they reached an enormous size. In France some were reared successfully on plum. According to a statement of my Spanish correspondent, Pernyi is essentially an oak feeder, which will degenerate after a time, if fed on other trees than oak.

Telea Polyphemus (North America).—This silkworm, which produces a closed cocoon. a little smaller than that of Polyphemus (North America).—This silkworm, which of a very superior quality. It can easily be bred like Pernyi in the open air, in England, unless the weather should be exceptionally bad. Polyphemus is now acclimatized in Spain, where I introduced it in 1879. In 1880, some 1,500 wild cocoons were collected from oak, birch, and other trees. It is very polyphagous. My Spanish correspondent in Spain, but he says a valuable acquisition to scrictculture in Spain, but he says a valuable acquisition to scrictculture in Spain, but he says a that a tendency to become double some Polyphemus as a valuable acquisition to scrictculture line Spain, but he says it has a tendency to become double some Polyphemus as a valuable acquisition to scrictculture line Spain, but he says it has a tendency to become double some Polyphemus as a valuable acquisition to scrictculture line Spain, but he says it has a tendency to become double some Polyphemus as a valuable acquisition to scrictculture line Spain, but he says it has a tendency to become double there, two male moths having emerged in November. The larve thrived well on birch (Edula alba).

The tree raine of the caterpillars, after a little experience will be found at the read of the larvæ. Cut lea

protect them from birds.

To give fresh food to larvæ reared on cut branches, as to protect them from birds.

To give fresh food to larvæ reared on cut branches kept in water, when the foliage has been eaten, or is too old and dry, is very easy. The old branches are merely placed in contact with fresh branches, or the old branches cut in pieces (not to be too heavy) are placed on the new ones. The larvæ, which should not be handled, will leave the old branches to go to the fresh ones. In a short time the old branches, bare of larvæ, may be removed.

When branches are plunged in a bottle, or any other vessel containing water, the foliage at the base of the branch should be cut off, as leaves in the water would decompose it, render the rest of the foliage unwholesome, and even poison the larvæ. The cut branches in water should be placed in the shade, where they will keep fresh for several days, especially if the foliage is sound and healthy, a condition of great importance. The water should be renewed, and the foliage freed of green flies and other small insects.

To rear Lepidoptera from the egg the moths abould be placed in cages for pairing and depositing their eggs. With moths of Sphingidas and some other species, it is useful to put in the cage a bunch of aromatic flowers, with branches of the plant the larvæ feed upon. Moisture should always be maintained in the cages.

A few days, or immediately after the eggs have been obtained, they should be placed under a glass with a small branch or leaves of the proper plants, so that the larvæ should find their food as soon as they are hatched.

When the larvæ are small, I rear them under bell-glasses, having a few holes on the dome. These glasses, which are of various sizes, according to the number of larvæ, rest on saucers full of sand covered with a piece of paper. Small branches of the proper food plants are stuck through the paper and plunged into the sand, where they keep fresh for several days without requiring any water.

The larvæ, under a bell-glass, can b

cages containing a few inches of light soil or soft sand, and this plan must always be adopted when the habit of the larve is not known.

Now, with respect to the sending of living cocoons and pupse from abroad, on the cases there should be written in large letters, "living pupse," or "living cocoons of silkworms," with request to keep the cases in the coolest place, or in the ice house of the vessel. The cocoons should be well packed in straw, hay, moss, or anything that will deaden the shocks to which the cases hay be subjected. Pupse o. lepidoptera must be placed in bran, sawdust, or fine moss. Cocoons and pupse should be sent as soon as possible after their formation, from the beginning of October to about the beginning of April, according to distance, so that they should not be subjected to the heat the whole of the time during their voyage to England. Small quantities of cocoons and pupse should be sent by sample post, in registered boxes, not exceeding the legal weight; the boxes must be strong, and it is best to the a label to each box, and affix the stamps to the label. Persons living too far inland to send living pupse may send dead specimens of the perfect insects (butterflies and moths). These should be in good condition, and placed with folded wings in paper envelopes. To protect these specimens from the attacks of mites, "Dermestes" beetles, and other parasites. It is important to put some poison in the boxes containing the specimens.

With respect to the sending of live cocoons and pupse, and even ova of lepidopters. I may say, that with a little care, and especially if they were given in charge of the captain, or some other person on board ship, they could be sent to Europe from distant countries, and arrive allve and in good condition.

In proof of this, I may mention the fact, that Mr. Youl, acting as agent of the Tasmanian Government, shipped, in

condition.

In proof of this, I may mention the fact, that Mr. Youl, acting as agent of the Tasmanian Government, shipped, in 1864, packed in a box, which was placed in the ice-house of the steamer Norfolk, 'a large quantity of salmon and trout ows, the result being the successful introduction of salmon and trout into the rivers of Tasmania and Australia.

In the same way, silkworm ova, live coccons, and pupe could safely be sent to Europe, from very distant countries, and this would be of the greatest interest and value to entomologists, for the study of lepidoptera in their various states.

To conclude, I shall reproduce the letter of one of my correspondents, Mr. J. P. Cock, whose death I accidentally learnt, on the 18th November last, in a house at Thames

<sup>•</sup> From a letter just received from Major G. Coussmaker, I hear that the article appeared in the Times of India, and also in the Indian Agriculturals.

Ditton, from Mr. P. Clarke, a gentleman who is a teaplanter in Assam. This sad news was recorded in an Indian paper, the Assam Gazette, of October 25, 1889, which, at my request, was forwarded to me a few days after.

I now give my correspondent's letter to me, dated 14th February, 1880, and received on the 12th of March, 1880:

KASSIA HILLS, ASSAM.

February, 1880, and received on the 12th of March, 1880:

Kassta Hills, Assam.

"Dear Sir: You must have thought it very remiss on my part, allowing your letter to remain so long unanswered, but a sudden and unforeseen calamity, in the death of my only brother, Major Cock, Deputy-Assistant Adjutant-General. Eastern Counties Districts, who fell mortally wounded while leading on his men in the final assault on Khonoma, in the Naga Hills, has entirely prevented me paying any attention to entomological pursuits for the last three months.

"My poor brother, having died possessed of a good deal of landed property in no less than three of our Indian hill stations, I have been traveling incessantly winding-up his affairs; in fact, I may with perfect truth say, that for the last two months and a half, I have been living in railway carriages and on board river steamers.

"The old adage, that misfortunes rarely come singly, I have found in my case to be true, for on my return to this station last Thursday, I found that my bungalow had been burnt to the ground through the gross carelessness of a drunken syce. Nothing was saved. A magnificent and most expensive library of entomological works, 47 large cabinets of specimens (my own private collections), my gleanings for over 26 years in Sumatra, Java, New Guinea, Borneo, Calebes, the Philippine Islands, and Japan, over 4,000 specimens ready to forward to England—all was lost just through the carelessness of a drunken wretch capsizing a lamp in my stables.

"I keep up a staff of eight Rhapias, whom I have thoroughly trained for the work of collecting in the malarious

through the carelessness of a drauds. It may stables.

"I keep up a staff of eight Rhapias, whom I have thoroughly trained for the work of collecting in the malarious jungles, where it is almost certain death for a European to aleep one night. I likewise have a large circle of friends and acquaintances among the officers and tea planters in the districts, all of whom I have persuaded to collect for me, and who send me monthly what they have been able to accumulate, and as I always take the field myself in March, and do not generally leave the forests before autumn is far advanced, many thousand insects pass through my hands annually.

late, and as I always take the field myself in March, and do not generally leave the forests before autumn is far advanced, many thousand insects pass through my hands annually.

"As before stated, all my large stock of preserved insects had been lost in the fire; however, I hope in the course of a month, or six weeks at the latest, to be able to dispatch you a first consignment. I will pay particular attention to your wishes about the cocoons of our various slik moths, and have already received letters from two intimate friends, who, perhaps, are two of the most eminent entomologists in India—Capt. Marshall and Col. Jones—both officers in the Royal Engineers. They inform me that they have written to some of their correspondents in other parts of the Himalayas to procure cocoons of such of the silk moths as are not procurable here. I can, however, promise to send you any number of cocoons of the following species: Attacus assumensis, Attacus allas, Actias selene, and Actias menas. Will you kindly write to me by first mail after the receipt of this, what cocoons do you consider most valuable, and the particulars that may be useful to me in forwarding them?

"I shall probably be away in the wilds of the Naga hill forests, but your letter will be forwarded on without delay. I should very much like to see some of your reports, it would give me very much pleasure to read them; and in return, will forward you a copy of my book on the genus Deslephila, which ought to be completed and published next month. It includes all the known Asiatic species of Cherocampa, Sphinx, Macrosila, Smerinthus, and the illustrations, over 400 in number, have taken me nearly three years to complete, as I have drawn each moth in water-colors as soon after capture as possible, with representations of the egg, caterpillar, and tree on which they live.

"As the season is not sufficiently advanced to take the field, hard frosts and bitterly cold winds prevailing at this lofty elevation, where anything in the shape of vegetation is parched and

The internation of the letter, to which I replied three times, I never received any communication from my correspondent, and, as above stated, it was by mere accident that I heard of his death.

Alfred Wailly.

110 Clapham road, London, S. W.

## THE MODE OF FLIGHT OF THE ALBATROSS.

force is the same. The bird directs his course mainly with his tail, the action of which upon the air is identical with the action of a ship's rudder upon the water. By this downward motion, his velocity rapidly increasing, he acquires a degree of momentum sufficient to carry him up again to a height equal to or greater than that from which he started. In this up and down long wave-like motion, with all its variations on either side, consists the whole of his flight day after day for hundreds of miles; at long irregular intervals he may give a few lazy flaps with his immense wings. Other birds use the mode of flight of the albatrose, but to a smaller extent, for the reason, in the case of smaller birds, that, the ratio of feathers to bulk being greater, their specific gravity is less, consequently they are unable to acquire the degree of momentum necessary to carry them upward; but on the other hand they have the power of sustained effort in moving their wings rapidly, which the albatrose has not. Gravitation then, which prevents him from rising directly on the wing, is the motive power of the albatrose when aloft. He must always take a run or paddle over the surface of the water in order to get a start, and on the land or the deck he is a prisoner, because he has no water in which to paddle himself along with his webbed feet, and he is unable to run. Instead of being assisted by the wind, his speed is lessened by just so much as the wind's velocity, when it happens that the direction of the wind and his intended course are opposed to each other, but with the wind his speed is just so much greater than it would be in a calm.

I do not advance this explanation as an imaginative theory. I claim more for it. I have had many opportunities of studying the movements of the albatross for consecutive days, and I feel confident that the above will be found to answer all required conditions.—Howard Sargent, in Nature.

#### ARCHÆOLOGICAL EXPLORATIONS NEAR MADI SONVILLE, OHIO.

THE valley of the Little Miami River, in southwestern Ohio, has been long noted for the number and extent of its prehistoric earthworks, which, distributed on both sides of the river, from its confluence with the Ohio to the well known Fort Ancient and beyond, form an almost continuous chain of mounds, forts, circles, and embankments, extending for more than fifty miles, and constituting an important division of the great earthwork system of the Mississippi Valley.

Valley. In October, 1878, Dr. Charles L. Metz contributed to the "Journal of the Cincinnati Society of Natural History" a



Section of Mound No. 5, Group A.

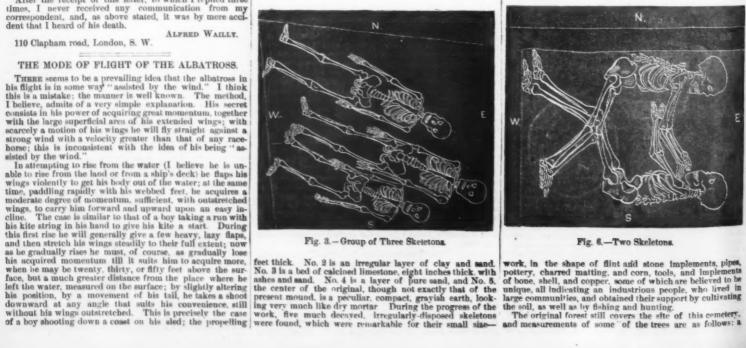
paper on the aboriginal remains of this vicinity, accompanied by a chart on which the mounds and earthworks were designated by symbols in accordance with the international code of MM. Mortillet and chantre. The examination and exploration of these remains, which was begun by Dr. Metz and a few other gentlemen, was prosecuted under the auspices of the Literary and Scientific Society of Madisonville, during the years 1878, 1879, and 1880, and the results are perhaps among the most interesting of any that have been conducted in the Mississippi Valley. The following



Fig. 2.—Section of Mound No. 6, Group A.

brief outline of the work and the discoveries that were made during the progress of it are condensed from the full report communicated to the Cincinnati Society of Natural History, and published in its "Journal" (vol. iii., numbers 1, 2, and 3).

1, 2, and 3).
In November, 1878, the earthwork known as Spice Bush Mound, was opened under the direction of Dr. Metz. The accompanying sketch (Fig. 1) shows the stratification of this mound, as seen in Section. No. 1 is a stratum of black leaf mould and gravelly clay, about one and a half to two



averaging but 5 feet 2 inches in length. All these were undoubtedly intrusive burials, and appear to have been thrown upon the original mound irregularly. A number of fragments of burned limestone, broken bowlders, a few film chips, and two small fragments of pottery were found in the mound.

e mound.

During the same month, another mound, situated on the

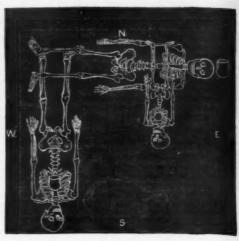


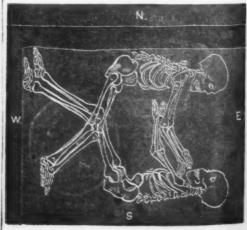
Fig. 4.—Group of Skeletons,

"Second Bottom," a plateau of the Little Miami River, about 150 feet above water line, and about one-third of a mile from the river, was explored. The section of this is shown in Fig. 2. In the center there is found a small circumscribed deposit of ashes, mixed with fragments of charcoal and charred bones, and with these were mingled fragments of a human skull, perfectly sound, so far as the action of fire was concerned, but soft and friable through decay. The excavated material was thrown back, and the mound restored,



Fig. 5.—Skeleton of Old Person.

as nearly as possible, to its original condition. In the month of March, 1879, during the progress of work on mound No. 5, group A, a laborer, while prospecting by digging holes in the surrounding forest, came upon a human skeleton at a depth of about two feet. This was the initiatory step toward a most important archæological discovery, as further investigation has revealed the fact that the entire plateau is the site of an ancient cemetery, from which have since been exhumed upward of four hundred skeletons of a prehistoric people, accompanied by numerous evidences of their handi-



& ALLUVIAL SOIL CHARRED WOOD & OTHER REMAINS OF FIRE ASHES & CLAY OR SAND 3. IN.
LEAR WHITE ASHES 2 IN
SAND ASHES & UNIO SHELL
6, IN.

Fig. 7 -Diagram of Ash Pit, No. 53.

over the surface; and it was until recently supposed to be a place where the manufacture of pottery had been carried on by the ancient inhabitants of the valley. The exploration of the cemetery, since its true character was discovered, has been, as far as the ground gone over is concerned, exceedingly thorough. Of the skeletons exhumed but a small proportion are in a good or even tolerable state of preservation; and the preservation of even these few must probably be ascribed to the favorable character of the soil—a compact



Fig. 8.—Ash Pit with Human Remains.

gravely drift—since the various surroundings, the position of some skeletons under large trees, etc., all indicate for these interments a remote antiquity.

The mode of burial seems to have been far from uniform. A large majority of the skeletons are found at a depth of two to three feet. in a horizontal position, face upward; but exceptions to this rule are numerous, many interments being made in a sitting posture, and some in groups of from three



Fig. 9.—Diagram of Double Corn Pit.

to six individuals regularly (Fig. 3) or irregularly disposed.

Rg. 5 shows the skeleton of a very old person, which was uncovered June 16th, 1879. Its position was partly extended, lying on its side, face east, with hands raised and knees pro-



Fig. 10.—Earthen Burial Vessel (Dr. Metz). One-third size.

tion of a stone coffin, but in one case the skeleton was cov-ered with a layer of small flat limestone from the adjacent stream. The heads of those in the horizontal position are generally directed to the east or southeast; but this rule is



Fig. 11.—Small Vessel (C. F. Low). One-half size.

not constant, several being found at right angles to these (Fig. 4). It is worthy of note, however, that with scarcely an exception, those skeletons which are accompanied by the finer vases, pipes, or other choice relics, have their heads

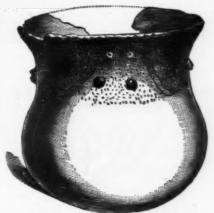




Fig. 13.—Earthen Vessel (Joseph Cox, Jr.). One-half size.

alout, 15½ feet in circumference; an oak, 12 feet; a maple, jected. The two skeletons shown in Fig. 6 were found di-feet; an elm, 13 feet. The locality has long been known, rectly over an ash pit; they were in semi extended positions, local archeologists as the "Pottery Field," so called on heads directed east and twest, and lower limbs crossed showing cheructeristic lesions strongly income of the numerous fragments of earthenware strewn. There has been no attempt in any instance at the construc-



Fig. 14.—Small Double Vessel (G. W. Lasher.) One-half size.

interest in its relation to the geographical distribution of the latter disease, and also as bearing on the theory of its introduction into Southern Europe from America in the fifteenth

century.

An interesting feature of these excavations has been the discovery of what may be designated as "ash pits;" being



Fig. 15.-Large Vessel (C. F. Low). One-third size.

circumscribed deposits of ashes, shells, sand, etc., from two to three feet in thickness, placed at varying distances below the surface. A perpendicular section made of one of these pits (Fig. 7) answers to the following description, which will serve to convey a fair idea of them all. Diameter of pit, 3 feet; the first 18 inches, of leaf mould and alluvial soil;



Fig. 16.—Earthen Vase (Joseph Cox. Jr.). One-fourth size.

then 9 inches of charred wood, burnt earth, and charcoal; next, 12 inches of ashes and animal remains; then 3 inches of clay or sand; next, 2 inches of white ashes; 8 inches of sand and Unio shells; and, finally, 12 inches of pure ashes. Total depth 5 feet 2 inches. These pits are quite uniform in size. Intermingled with the ashes are pipes, implements



Fra. 17.

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Fig. 18.—Earthen Vessel. One-third size.

vertebra only. The discovery of this skeleton furnishes some clew to the purposes of the pits, and favors the view that they were probably places for temporary burial, from which the remains were afterwards removed for interment



Fig. 19.—Vessel with Salamander Ornamentation. One-third size.

in some of the numerous sepulchral tumuli usually called "battle mounds," or "sacrificial mounds." That these pits are very ancient is evident from the fact that subsequent



Fig. 20. - Vessel with Salamander Handles. One-third size

interments, in both sitting and horizontal positions, have been made directly over these excavations since the removal of the human remains, and forest trees of several hundred years' growth are now growing over these comparatively



Fig. 21.—(W. C. Rogers' Collection.) Inscribed Stone from Ancient Cemetery, Madisonville, Ohio.

of the pits, opened January 20th, 1890, there was found at the bottom an entire human skeleton (Fig. 8); but this is an A correct diagram of the pit (Fig. 9) is here given, along exception to the rule, since in no other case but one were human remains discovered, and that was a single dorsal clay, in which were numerous animal remains, several im-



Fig. 22.—Stone Pipe (R. O. Collis)

plements of flint, stone, and bone, an unfinished pipe, and some charred bones. Third, 10 inches of ashes, intermingled with the bones of a great number of animals. Fourth, about 5 inches of coarse matting and twigs, cornstalks and bark, all completely carbonized. Fifth, a layer of shelled corn, probably 3 or 4 bushels, and below this a quantity of ear corn, all of which was completely carbonized. The bottom of the pit was covered with a layer of fire cracked bowlders, some ashes, and a few animal bones. The adjoining pit was separated from the corn pit at the bottom by about six inches of clay, and did not differ from the usual pits, except that no implements were found



-Stone Pipe (Joseph Cox, Jr.).

in it. From the uncharred condition of the articles usually found in the ash pits it is evident that the ashes have been placed in the pits as ashes, after having been burned elsewhere, as in no case do the walls of the pit show any traces of the action of fire.

The pottery ware which accompanies the skeletons is usually found situated near the head, and present many features of special interest. It is made of clay, finely tempered with powdered Unio shells, and much care has evidently been bestowed on its manufacture, some pieces being scarcely thicker than an ordinary teacup. Many specimens are in a perfect condition, or nearly so, and they usually contain a



Fig. 94.—Stone Pipe (C. F. Low).

single Unio shell when found, the shell being evidently intended for use as a spoon. The vessels range in capacity from a third of a pint, or even less, up to a gallon or more—the smaller ones being found in the graves of children. They are symmetrical in shape and varied in design, some being artistically ornamented with scroll work, some with handles representing lizards, human heads, etc., and they are almost universally provided with four handles. Fig. 10 shows an earthen vessel; found in a fine state of preservation, near a skull, in March, 1879. Fig. 11 represents a vessel found in an ash pit April, 1879. Fragments of two other vessels were found in the same excavation. On May 7th, of the same year, along with a skeleton six feet in length, was found the



Fig. 25.—Catlinite Pipe (E. A. Conkling).

Fig. 14 represents a small, two-story vessel, which was exhumed from a grave containing seven skeletons, two of which were those of children.

Fig. 15 represents a fine, perfect vessel, of about one gallon capacity, found at the feet of one of a group of skeletons, in the month of July.



Fig. 26.-Stone Pipe (G. W. Lasher).

Fig. 16 represents a vessel which was taken from the right side of the head of a male skeleton, October, 1879. It is provided with a base or pedestal, and is the only one of this peculiar form which has as yet been discovered.

In Fig. 17 is shown a peculiar narrow-necked vessel which was found under a skull in one of the ashpits, November 27, 1879. Originally it had four handles, as shown by the dotted lines.

On February 10, 1880, in excavating in what was supposed to be a hearth or irregular ashpit, it became evident to the explorers that the place was a kitchen-midden. The leaf mould was of about the same depth as in other parts of



Fig. 27.—Catlinite Pipe (R. O. Collis),

the cemetery, and several skeletons were found within the space. Along with a group of five of these was discovered the broken vessel, ornamented with a human face at the left side, seen in Fig. 18.

Fig. 19 represents a vessel which was found crushed between two skulls, but which was afterwards carefully restored from the fragments. The salamander-like ornamentation of this vessel is entirely new and peculiar to this cemetery. Several fragments and handles of other vessels have been found representing four or five species of the Salamandridæ or other Urodelæ.

With the remains of one of three skeletons exhumed July



Fig. 28.—Stone Pipe (A. A. Hawes).

4, were found two bone beads, a large number of copper beads, and the small, perfect vessel, with salamander ornamentation, shown in Fig. 20. This vessel was found near the lower extremities of the skeleton, which was that of an adult female.

There is a good reason to believe that each interment was criginally accompanied by a vessel although at present there

adult female.

There is a good reason to believe that each interment was originally accompanied by a vessel, although at present there is a great disparity between the number of these and of the skeletons found. This is perhaps to be accounted for by the fragments thickly strewn over the surface and inter-



Fig. 29.-Pipe (P. G. Thomson).

Fig. 21.—(W. C. Rogers' Collection.) Inscribed Stone from Ancient Cemetery, Madisonville, Ohio.

finely shaped ornamented vessel shown in Fig. 12. In lieu of the usual handles there were two perforations on each side and two small projections on each quarter.

The curious, gourd shaped earthen vessel shown in Fig. 22. In lieu of the missing burial uras. Among the other articles of utility or ornament found in the graves are pipes of various patterns, several of them askeleton, in connection with these ash pits, was made August 26th, 1879. In excavating a pit, a large

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heads, and many ornaments and implements of bone, such as beads, awis, needles, perforated teeth, etc., together with capacity, and two rough stone axes or fleshers.

Fig. 25 represents a finely-finished pipe, of curious form, made of dark red callinite. It was found with five skelenow in the collection of Mr. F. W. Langdon, is an irregular tons, all of which were in a horizontal position.

Forated shell disks about the size of a silver dellar, and a capacity, and two rough stone axes or fleshers.

Fig. 25 represents a finely-finished pipe, of curious form, made of dark red callinite. It was found with five skeleton.

Fig. 37 shows a shell ornament and bead, which were



Fig. 30.—Limestone Pipe.

piece of sandstone, measuring about 3×2×1 inches, on the flat surface of which are cut two parallel figures made of straight lines and apparently intended to represent arrows. The second stone, in the collection of E. A. Conkling, is a flattened dark-green bowlder measuring about 3½×2½ inches, one side of which is covered with a network of lines from ½ to ¾ of an inch apart and crossing each other at



Fig. 31.-Sandstone Pipe.

nearly right angles, thus forming quadrangular divisions of various sizes. The third (Fig. 21) was found in an ashpit in September. 1879. It is an irregular piece of fossiliferous limestone of a reddish-brown color, as though it had been stained by being deposited in a ferruginous soil, the fracture on the edge showing the natural color of the limestone. The markings are incised lines and the pointer is the most



Fig. 32.-Limestone Pipe,

prominent figure; the other lines are plainly visible, although the surface is much weathered and worn. The stone and markings, perhaps, have reference to the pit of carbonized maize, near which it was fcund, and it is to be regretted that the exact position in which it originally lay was not



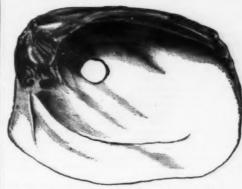


Fig. 84.—Perforated Unio Valve. One-third size.

Fig. 26 shows an ornamented stone pipe, found with an earthen vessel in an excavation made in April, 1879. This is now in the collection of Mr. G. W. Lasher.

Fig. 27 is a finely polished callinite pipe, which was found near the vessel numbered 12, above.

Fig. 28 represents a pipe made of limestone, which was discovered during one of the May excavations, lying near the cranium of a skeleton.

With a skeleton exhumed in October, 1879, was found, at



Fig. 35.-Shell Ornament (D. S. Hosbrook).

the right of the head, a broken vessel, and on the left side a pipe, Fig. 29, made of limestone, and carved to represent the head of some animal. A copper ornament, Fig. 40, was also found at the right side of the neck.

Fig. 30 represents a pipe of peculiar form, made of limestone, which was taken from an ashpit, April, 1880.

Fig. 31 represents a small sandstone pipe, which was picked up on the surface, at about fifty feet from where an excavation was being made in May. This relic is ornamented with rude carvings, representing a bird with out-

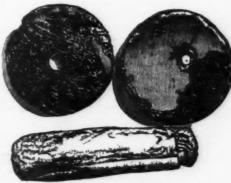


Fig. 36.—Shell Ornaments (Mr. Ferris).

stretched wings, and had been rooted up by the hogs which had been turned loose in the woods.

Fig. 32 represents another pipe, which was also found on the surface. The material is limestone.

Fig. 38 shows a sandstone pipe, which was found in June in an ashpit, along with two rolls of copper, five bone beads, one ungrooved ax, and a stone dresser.

Numerous perforated implements and ornaments made from the shell of Unio, were found in the ashpits, and of which Fig. 34 is an illustration of the largest one discovered.





found in May, 1880, near the remains of an adult female and

found in May, 1880, near the remains of an adult female and a child.

Fig. 38 represents a very interesting object, perhaps a rattle, which was found with the skeleton of a child in September, 1879. It is made of a single piece of copper of irregular shape, the edges of which have been brought together so as to form a bell, or rather what looks like a sleigh bell, leaving an irregular opening on one side. A small hole was punched through the top and a strip of copper doubled up and the ends pushed through the opening from the inside so as to form a handle. Inside this bell is a fragment of copper, about the size of a large pea, and when the object is shaken it produces a rattling or tinkling sound.



Fig. 39.-Elk Horn Ornament (E. A. Coukling).

It is without question one of the most unique specimens of aboriginal workmanship ever recovered.

From an ashpit opened in October, 1879, was taken the singular ornament represented in Fig. 39. It is made of elkhorn, and may possibly have been utilized as a comb. The copper ornament shown in Fig. 40 has already been spoken of above.

As regards the particular race to which the people belonged whose remains are found in this extensive cemetery—whether they were identical with or related to the celebrated "stone-grave people" of Tennessee, as some of their potteryware and the shape and dimensions of their



Fig. 40.—Copper Ornament (H. B. Whetsel).

crania would seem to indicate; or whether they were the last remnants of the once powerful nation that erected Fort Ancient and other gigantic works in this region—these and similar queries remain as yet unanswered. More extended investigations and a careful comparison of large amounts of material from this and other localities may be expected to assist in the solution of these obscure but interesting questions.

### THE GRAPES OF CALIFORNIA.

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Fig. 33.—Shell Ornament and Bead.

Fig. 35.—Shell Ornament, which was found with a vessel, near a skeleton in one of the pits.

Fig. 35.—Shell Ornament, which was found with a vessel, near a skeleton in one of the pits.

Fig. 35.—Shell Ornament and Bead.

Fig. 36.—Shell Ornament and Bead.

Fig. 37.—Shell Ornament and Bead.

Fig. 38.—Sandstone Pipe.

Fig. 38.—Sandstone Pip

teen tone per acre. The vineyards are along the coast, upon the foot-bills of the Sierras, in the broad valley that forms the central part of the State and on the mountain sides to the east. The native California vine called the "Mission" is a different species from the varieties grown in Europe. The California vine is hardier than the European, and the wine it produces is not very palatable. Cuttime, grafts, and seed were on this account obtained from Missouri, Ohio, and Europe as early as 1853. The Mission grape, however, continues to be cultivated more than all other varieties, and to secure the best product from its culture the practice of "blending" it with finer varieties grave the same continues to be cultivated more than all other varieties, and to secure the best product from its culture the practice of "blending" it with finer varieties grave the same factors. Professor E. W. Hilgard aya: "Judicious blending has become the height of the art of wine-making, for it certainty is an art, and a difficult one, to produce the best result from materials so infinitely varied, even as regards the same grape variety in different seasons. So far the blending has been chiefly in the hands of the wine merchants of San Francisco. It is no reflection upon these gentlemen to say that they have not been allogether successful in their efforts to adapt the uncertain raw wines that come to their cellars to the established tastes of the world's market. It stands greatly to their credit that on the whole neutral spirits, log wood, glycerine, and sulphyric need thave played but a small formia wines for purity—that is, for containing only the juice of the grape—bas not suffered at their hands."

Commissioner C A, Wetmore mentions nine species of American vines, the Rupestris, Cordifolia, Riparia, Arizonica, Californica, Æstivalis, Candicaus, Labrucca, and Vulpina. Of these there are over one hundred varieties. The Californic flourishes everywhere, and may be grown from the seed with difficulty. Seeding the played the played t

drinkers, as a pint of wine costs no more than a cup of tea or coffee."

A chapter is devoted to the raisin-making industry. It is only within the last few years that a marketable raisin has been produced. In 1879 75,000 boxes were produced. The product of the State is now greater than the consumption, and no raisins are imported to the State. The California product is yearly improving in quality, and there is every reason to believe that a large export trade to the Eastern States will grow up in a few years.

There are still many things standing in the way of rapid improvement of quality in California wines and the development of the industry upon a solid foundation. The higher price of labor is a great disadvantage. The vines, as a rule, are not pruned close enough and are allowed to bear too many bunches. California growers are importing vines from Missouri, Ohio, and Europe; while France, Spain, Portugal, Italy, and Australia are sending in orders for California vines. There is a great dearth of experts in viticulture and the manufacture of wine. Too little attention is paid to fertilizers. The bone-meal factories of San Francisco send all their products to New Zealand and Australia, Yet in a few years it is predicted the wine product of California will exceed 100,000,000 gallons. The State is the only one where the European varieties can be successfully grown. The State now consumes herself three-fourths as much as imported into the United States from all foreign

countries, and exports to the Atlantic States more than the whole country imports from France. As Commissioner Rose says: "The possibilities here are immense. A great future is in store for us, if it is a fact, and I believe it, namely, that Europe will buy our wines."

BUTTER AND BUTTER-MAKING.\*

DR. E. L. STURTEVANT, of Framingham, was the first speaker. He commenced with the axiom: Milk was the basis of butter-making. One parcel of milk, he said, would make butter, while another would not, showing great difference in its characteristics. The quality of milk may be influenced by the feed given to the cows. He would treat of these animals as butter cattle, stock cattle, and store cattle, confining his remarks at present to the former. He raised four tons of corn stover to the acre, equal in feeding value to six-tenths the weight of good hay. Good butter cannot be made from the milk of cows feed entirely on corn stover; in fact, milk is deteriorated to the extent of stover thus fed. Corn-meal does not increase the flow of milk. Give a cow all the good hay she will cat, and the addition of corn-meal will not increase the flow of milk. In conclusion, Henry E. Alvord made some brief remarks

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A milk yielding eight per cent, of cream will often give more butter than one yielding sixty per cent. The constituents of milk can best be illustrated by the use of the microscope in comparing the clusters of butter-globules therein contained. Through this test the Jerseys were proved emphalically, teaving a blue skim-milk. Lokes of a whitish color, smallness of globules is indicated, and an imperfect separation from the milk. The length of time consumed in churning is determined largely by the size of the butter-globules, while corn-meal tends to increase their size, and hence is best for butter. Shorts are the poorest feed known in butter-making, reducing the size of the butter-globules, while corn-meal tends to increase their size, and hence is best for butter. Shorts are the poorest feed known in butter-making, reducing the size of the globules. Cotton-seed meal conveys a bad taste to butter, reducing the market value of the product. Linseed-meal occasions no injury, while cotton-seed for his produced the formal p

washed with iced brine until the buttermilk is pretty well washed out. He uses a butter-worker, never allowing the butter to be touched with human hands. Finally, to get out all the moisture, he uses large sponges, with which he rolls up the butter after having been rolled out in the worker, continually mopping the flattened butter. The sponges are kept in ice-water. He rolls the butter out three or four times, according to judgment; uses half an ounce of salt to the pound of butter; keeps the product covered, to prevent the escape of the aroma into the air; churns the cream one day, putting the butter product into balls or lumps the next day, usually making half-pound lumps. His butter retails in Boston at eighty cents per pound.

Jersey and Guernsey dairy cows are kept in about equal numbers; turns his cows out to grass in the summer season; takes two weeks in effecting, in a gradual way, the complete change from hay to grass; feeds his cows what they need before watering; experimented by feeding one cow with eight quarts of shorts per week, being obliged, in consequence, to discount on her butter product when it reached the market. Good butter cannot be obtained if cows are fed with shorts. He cuts his hay in June; allows his milk to stand twenty-four hours to raise the cream; and thinks the system of deep setting gives him increased quantity of butter.

Mr. Bowditch continued by stating that Franklin County

his milk to stand twenty-tour notified raise the cream, and thinks the system of deep setting gives him increased quantity of butter.

Mr. Bowditch continued by stating that Franklin County usually sends 838,000 pounds of butter to Boston market. An increase in price of 10 cents per pound would make an extra profit of \$80,000 to this county alone. The average price of Franklin County butter was but 25 cents per pound; a single county in Vermont sends butter to Boston that averages 40 cents per pound; choice Western creamery butter commands 37 cents per pound. Making butter at 25 cents a pound is not a profitable business, and there is no necessity for selling fine butter at that price.

In answer to a query why the human hand injured the butter, the speaker remarked that there are innumerable pores in the skin, which throw off the wastes of the system. The cleaner the hand, the less the obstruction to this process and the easier the refuse matter of the body, or the insensible perspirations, was disposed of. All such matters injuriously affect the purity of the butter, to say nothing of the \*Abstract of addresses at a meeting of the Franklin County Institute,

\* Abstract of addresses at a meeting of the Franklin County Inst held at Greenfield, Mass., March 19, 1861.

quarts of milk.

quarts of milk.

In conclusion, Henry E. Alvord made some brief remarks upon the proper method of judging of the quality of butter as exhibited at our fairs. Often it is the case that the reputation or popularity of the individual who owned the butter drew the premium, instead of the excellence of the butter product on exhibition. He wished for the adoption of a system of marking for flavor, grain, color, salting, and style, say on a scale of one hundred points, a plan which the managers of our coming fall fairs might do well to consider and carry into execution.—Amer, Cultivator.

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